



हिमाचल प्रदेश तकनीकी विश्वविद्यालय

हमीरपुर-177001, हिमाचल प्रदेश (भारत)

[हिमाचल प्रदेश अधिनियम-16, 2010 के अधीन स्थापित]

HIMACHAL PRADESH TECHNICAL UNIVERSITY

HAMIRPUR – 177001, HIMACHAL PRADESH (INDIA)

[ESTABLISHED UNDER STATE LAGISLATIVE ACT-16 OF 2010]

सूचना पत्रिका

INFORMATION BROCHURE

Himachal Pradesh Common Entrance Test

(HPCET – 2016)

PART-I

(FOR ADMISSION TO VARIOUS TECHNICAL & PROFESSIONAL COURSES)

IMPORTANT INSTRUCTIONS

1. The HPCET-2016 Application Form has been made completely online. The candidates have to fill the particulars online and also upload their photograph and signature. The candidates must send a scanned copy of the Application Form duly countersigned by their parents through e-mail on HPTU email-id himtuadmission@gmail.com on or before 2nd May, 2016. The candidates are further advised to retain hard copy of the Application Form for future reference/correspondence, if any.
2. Entrance Examination Fee is to be paid either by credit/debit card or through e-challan generated during the online filling of the Application Form. In case of e-challan, the payment should be made in the Punjab National Bank in cash. Fee once paid will not be refunded (full or partial) under any circumstances.
3. Enter complete address giving your name and Pin-code carefully. The name of the candidate and his/her parent's name in the Application Form should exactly be the same as registered in Class 10th/equivalent qualifying examination. No prefix/title such as Mr/Shri/Dr/Mrs/Smt. etc. should be used.
4. The use of address of any Coaching Centre/Institution is strictly prohibited. No options can be changed at later stage after submission of an Application Form.
5. No change will be accepted through offline i.e. through fax/application including e-mail etc.
6. You must quote your Application Form number as reference in all your future correspondence.
7. The decision of the HPTU regarding the eligibility of any applicant shall be final.
8. For latest information related to HPCET-2016, please visit University Website: www.himtu.ac.in

IMPORTANT DATES

HPCET-2016

Programme	Start of filling up Online Application Form	Closing of filling up Online Application Form	Date of Entrance Examination
MBA	02-03-2016	01-05-2016	14.05.2016 (10 AM- 12Noon)
MCA	02-03-2016	01-05-2016	14.05.2016 (2 PM- 4PM)
B. Tech (Direct Entry), B. Pharmacy and B. Pharmacy (Ayurveda)	02-03-2016	01-05-2016	14.05.2016 (10 AM- 1PM)
M.Tech	02-03-2016	01-05-2016	15.05.2016 (2 PM- 5PM)
M.Pharmacy	02-03-2016	01-05-2016	

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1. ABOUT THE UNIVERSITY AND ADMISSION TO TECHNICAL AND PROFESSIONAL COURSES

- (a) The Himachal Pradesh Technical University, Hamirpur, a premier institution of undergraduate, postgraduate teaching and researching the country, was established by an Act of Legislative Assembly (Act No. 16 of 2010) on July 30,2010 as a State Government University of Himachal Pradesh.
- (b) Under the provision of Section 5 of the Himachal Pradesh Private Technical and Vocational Educational Institutions (Regulation of Admission and Fixation Fee) Act, 2008, the State Government vide its Notification No. EDN(TE)F(10 2/2013 dated 13th August, 2015 has notified the eligibility criteria for admission in Technical and Professional Courses from academic session 2016-17 onwards in respect of the institutions as specified under Section 2 of the Act 2008.
- (c) In compliance to above referred notification Himachal Pradesh Technical University (hereinafter called as HPTU) will conduct Common Entrance Test (hereinafter called HPCET-2016) for admissions to all Technical, Vocational or Professional Courses in all the institutions deemed to be University, or Universities established under the State Act or Constituent units thereto.
- (d) The University will conduct the HPCET-2016 for admission to B.Tech./B.Pharmacy/ B.Pharmacy (Ayurveda), M.Tech, M.Pharmacy, MCA and MBA being offered by HPTU and colleges affiliated to it, deemed to be University, or other Universities established under the State Act or Constituent units thereto.
- (e) The admission to all courses shall be made on the basis of merit or rank/marks obtained in the National Level Entrance Test subject to fulfillment of minimum educational qualification. However after exhausting the merit of National Level Entrance Test, the admission to all courses shall be made on the basis of merit of marks obtained in HPCET-2016.

2. PATTERN OF THE HIMACHAL PRADESH COMMON ENTRANCE TESTS-2016 (HPCET-2016)

2.1 MODE OF ENTRANCE TESTS

- (a) The Entrance Test for each course shall be conducted in single offline (Pen and Paper Based Test) mode.
- (b) All questions shall be objective type (multiple choice questions); each question with choice of four answers. The candidate will have to choose one correct answer only.
- (c) Two marks shall be awarded for each correct answer and 0.5 mark shall be deducted for each wrong answer.

- (d) While the candidate can skip a question, but should not choose more than one option as correct answer.
- (e) Subject combinations for Entrance Test, number of questions and mode of test for different courses is as below:

(i) B.Tech/B.Pharmacy/ B.Pharmacy (Ayurveda)

Section	Subjects	No. of Questions	Max. Marks	Duration
A	Physics	50	100	3 Hrs
B	Chemistry	50	100	
C	Mathematics/Biology	50	100	
Total		150	300	

Note: Candidates attempting Physics, Chemistry & Mathematics shall be eligible for B.Tech/B. Pharmacy whereas candidates attempting Physics, Chemistry & Biology shall be eligible for B. Pharmacy or B. Pharmacy (Ayurveda).

(ii) M. Tech

Section	Subjects	No. of Questions	Max. Marks	Duration
A	Aptitude Test	15	30	3 Hrs
B	Civil Engg./Electrical Engg./Electronics & Comm. Engg./Computer Science & Engg./Mechanical Engg.	60	120	
C	Engg. Mathematics	25	50	
Total		100	200	

Note: A candidate seeking admission in M.Tech is expected to appear in a paper appropriate to the discipline of his/her qualifying degree of section B above. The candidate is, however, free to choose any paper according to his/her admission plan, keeping in mind the eligibility criteria of the institutions in which he/she wishes to seek admission.

(iii) **M. Pharmacy**

Section	Subjects	No. of Questions	Max. Marks	Duration
A	Aptitude Test	15	30	3 Hrs
B	Pharmaceutical Sciences	60	120	
C	Bio-Chemistry, Microbiology, Pharmaceutical Jurisprudence	25	50	
	Total	100	200	

(iv) **MCA**

Section	Subjects	No. of Questions	Max. Marks	Duration
A	Verbal Ability	25	50	2Hrs
B	Quantitative Ability	25	50	
C	Data Interpretation and Reasoning	25	50	
D	General Knowledge and Computer Awareness	25	50	
	Total	100	200	

(v) **MBA**

Section	Subjects	No. of Questions	Max. Marks	Duration
A	Verbal Ability	25	50	2 Hrs
B	Quantitative Ability	25	50	
C	Data Interpretation and Reasoning	25	50	
D	General Knowledge and Business Awareness	25	50	
	Total	100	200	

2.2 SYLLABUS FOR ENTRANCE EXAMINATION

The syllabus for appearing in different tests is given in Appendix (A to C) as indicated against the name of course below:

Sr. No.	Name of Course	Appendix
1.	B.Tech	A
2.	B. Pharmacy	
3.	B. Pharmacy (Ayurveda)	
4.	M.Tech	B
5.	M. Pharma	C
6.	MCA	D
7.	MBA	

2.3 LANGUAGE OF THE QUESTION PAPERS

The language of all question papers will be English.

3. ELIGIBILITY CRITERIA

The candidates appearing in HPCET-2016 for seeking admissions to a particular course must fulfill the eligibility criteria for the corresponding course as per AICTE Norms. The minimum eligibility criterion for different courses is given in Table (1).

Table (1): Minimum eligibility criteria for admission to different programmes/courses as per AICTE Norms

Name of Programme/ Course	Minimum Eligibility Criteria
B.Tech. (Direct Entry)	Passed 10+2 or its equivalent examination from a recognized Board or University with Physics and Mathematics as compulsory subjects along with one of the Chemistry/Biotechnology/Biology/Technical Vocational subject securing at least 45% marks (40% in case of candidate belonging to reserved category) in the above subjects taken together.
B.Pharm (Direct Entry)	Passed 10+2 examination with Physics and Chemistry as compulsory subjects along with one of the Mathematics/Biotechnology/Biology/ Technical Vocational subject securing at least 45% marks (40% in case of candidate belonging to reserved category) in the above subjects taken together. Provided that a student should complete the age of 17 years on or before 31 st December of the year of admission to the course.
B.Pharm (Ayurveda)	Passed 10+2 examination from a recognized Board with Physics, Chemistry and Biology as compulsory subjects with at least 45% marks (40% in case of candidate belonging to reserved category) in the above subjects taken together.
M. Tech	Recognized Bachelor's degree or equivalent in the relevant field having at least 50% marks (45% marks in case of candidate belonging to reserved category) .
M. Pharmacy	Recognized Bachelor Degree in Pharmacy or equivalent having at least 50% marks (45% marks in case of candidate belonging to reserved category).
MBA	Recognized Bachelor's degree of minimum three years duration with at least 50% marks (45% marks in case of candidate belonging to reserved category).
MCA	Recognized Bachelor's degree of minimum three years duration with Mathematics at 10+2 level or at Graduate level securing at least 50% marks (45% marks in case of candidate belonging to reserved category).

- Note:**
- i) The candidates who are appearing for their final examination of 10+2/Bachelor Degree examination in March/April, 2016 shall also be eligible to appear in HPCET-2016. However the final selection shall be subject to satisfying the above minimum eligibility criteria.
 - ii) In case the percentage in aggregate comes in fraction then the percentage of minimum marks obtained in qualifying examination as required for admission to various courses mentioned in the Admission Brochure of HPTU, shall be rounded off to the nearest percentage of marks e.g. 44.50 – 44.99 % shall be rounded off to 45% and 44.01 – 44.49% shall be rounded off to 44%.

4. ADMISSION CRITERIA

- (a) Admission to the above programmes will be made strictly on the basis of merit or rank/marks obtained in the National Level Entrance Test/HPCET-2016 subject to fulfillment of minimum educational qualification as stated in Table (1).
- (b) The criteria for making admissions to various Under Graduate and Post Graduate Programmes as per norms notified by the Government of Himachal Pradesh is given in Table (2).

Table (2): Admission Criteria

Name of Programmes/ Course	Admission Criteria												
B.Tech. (Direct Entry)	<ol style="list-style-type: none"> 1. The admission to the first 50% of the seats shall be made on the basis of merit/rank/score obtained in JEE (Main)-2016. The remaining 50% seats will be filled up on the basis of merit of HPCET-2016. 2. In case the seats remain vacant after exhausting the merit of JEE (Main) and HPCET, the remaining seats will be filled up on the basis of merit of 10+2 examination conducted by a recognized Board or University. 3. The number of seats remaining vacant after making admission on the basis of each test/examination will be shifted and filled on the basis of next test/examination in continuous order. <p>Note:- The minimum/cut off marks to seek admission to B.Tech degree course on the basis of JEE (Main)-2016, HPCET-2016 and 10+2 shall be as follows:</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="text-align: center;">Name of Exam</th> <th style="text-align: center;">General Category</th> <th style="text-align: center;">Reserved Categories</th> </tr> </thead> <tbody> <tr> <td style="text-align: center;">JEE (Main)-2016</td> <td style="text-align: center;">15%</td> <td style="text-align: center;">10%</td> </tr> <tr> <td style="text-align: center;">HPCET-2016</td> <td style="text-align: center;">20%</td> <td style="text-align: center;">15%</td> </tr> <tr> <td style="text-align: center;">10+2</td> <td style="text-align: center;">55%</td> <td style="text-align: center;">50%</td> </tr> </tbody> </table> <p>Wherein the % indicates the marks obtained by the candidate in the entrance test/10+2 examination out of the total marks assigned to the entrance test/examination.</p>	Name of Exam	General Category	Reserved Categories	JEE (Main)-2016	15%	10%	HPCET-2016	20%	15%	10+2	55%	50%
Name of Exam	General Category	Reserved Categories											
JEE (Main)-2016	15%	10%											
HPCET-2016	20%	15%											
10+2	55%	50%											

B.Pharmacy (Direct Entry)	<p>1. The admission to B. Pharmacy first year degree course shall be made on the basis of merit of HPCET-2016.</p> <p>2. In case the seats remain vacant after exhausting the merit of HPCET-2016 the remaining seats will be filled up on the basis of merit of 10+2 examination conducted by a recognized Board/University.</p> <p>Note:- The minimum/cut off marks to seek admission to B. Pharmacy degree course on the basis of HPCET-2016 and 10+2 shall be as follows:</p> <table border="1" data-bbox="524 390 1474 506"> <thead> <tr> <th>Name of Exam</th> <th>General Category</th> <th>Reserved Categories</th> </tr> </thead> <tbody> <tr> <td>HPCET-2016</td> <td>20%</td> <td>15%</td> </tr> <tr> <td>10+2</td> <td>55%</td> <td>50%</td> </tr> </tbody> </table> <p>Wherein the % indicates the marks obtained by the candidate in the entrance test/10+2 examination out of the total marks assigned to the entrance test/examination.</p>	Name of Exam	General Category	Reserved Categories	HPCET-2016	20%	15%	10+2	55%	50%
Name of Exam	General Category	Reserved Categories								
HPCET-2016	20%	15%								
10+2	55%	50%								
B.Pharm (Ayurveda)	The admission to B. Pharmacy (Ayurveda) first year degree course shall be made on the basis of merit of HPCET-2016.									
M. Tech	<p>1. The admission to M. Tech degree courses shall be made on the basis of merit of valid GATE-2016 score.</p> <p>2. In case the seats remain vacant after exhausting the merit of GATE, the vacant seats will be filled up on the basis of merit of HPCET–2016.</p> <p>3. In case the seats remain vacant after the exhausting the merit of National Level Test and HPCET-2016, the remaining seats shall be filled up on the basis of merit of B.E./B.Tech degree awarded by a recognised University.</p> <p>Note:- The minimum/cut off marks to seek admission to M. Tech degree course on the basis of HPCET–2016 and B.E./B. Tech. examinations shall be as follows:</p> <table border="1" data-bbox="524 1108 1474 1224"> <thead> <tr> <th>Name of Exam</th> <th>General Category</th> <th>Reserved Categories</th> </tr> </thead> <tbody> <tr> <td>HPCET-2016</td> <td>20%</td> <td>15%</td> </tr> <tr> <td>B.E./B. Tech</td> <td>60%</td> <td>55%</td> </tr> </tbody> </table> <p>Wherein the % indicates the marks obtained by the candidate in the entrance test/B.E./B.Tech examination out of the total marks assigned to the entrance test/examination.</p>	Name of Exam	General Category	Reserved Categories	HPCET-2016	20%	15%	B.E./B. Tech	60%	55%
Name of Exam	General Category	Reserved Categories								
HPCET-2016	20%	15%								
B.E./B. Tech	60%	55%								
M.Pharmacy	<p>1. The admission to M. Pharmacy degree courses shall be made on the basis of merit of Valid GPAT-2016 score.</p> <p>2. In case the seats remain vacant after exhausting the merit of GPAT, the vacant seats will be filled up on the basis of merit of HPCET-2016.</p> <p>3. In case the seats remain vacant after the exhausting the merit of National Level Test and HPCET-2016, the remaining seats shall be filled up on the basis of merit of B.Pharmacy degree awarded by a recognised University.</p> <p>Note:- The minimum/cut off marks to seek admission to M. Pharmacy degree course on the basis of HPCET–2016 and B. Pharmacy examinations shall be as follows:</p> <table border="1" data-bbox="524 1766 1474 1881"> <thead> <tr> <th>Name of Exam</th> <th>General Category</th> <th>Reserved Categories</th> </tr> </thead> <tbody> <tr> <td>HPCET–2016</td> <td>20%</td> <td>15%</td> </tr> <tr> <td>B. Pharmacy</td> <td>60%</td> <td>55%</td> </tr> </tbody> </table>	Name of Exam	General Category	Reserved Categories	HPCET–2016	20%	15%	B. Pharmacy	60%	55%
Name of Exam	General Category	Reserved Categories								
HPCET–2016	20%	15%								
B. Pharmacy	60%	55%								

	<p>Wherein the % indicates the marks obtained by the candidate in the entrance test/B. Pharmacy examination out of the total marks assigned to the entrance test/examination.</p>												
MBA	<p>1. The admission to M.B.A. degree course shall be made in the order of merit of Valid Score of National Level Tests CAT, CMAT, MAT etc. respectively.</p> <p>2. In case the seats remain vacant after exhausting the merit of National Level Tests, the same shall be filled up on the basis of merit of HPCET-2016.</p> <p>3. In case the seats remain vacant after the exhausting the merit of National Level Tests and HPCET-2016, the remaining seats will be filled up on the basis of merit of Bachelor degree examination conducted by a recognised University.</p> <p>Note:- The minimum/cut off marks to seek admission to MBA degree course on the basis of National, HPCET–2016 and Bachelor degree level examinations shall be as follows:</p> <table border="1"> <thead> <tr> <th>Name of Exam</th> <th>General Category</th> <th>Reserved Categories</th> </tr> </thead> <tbody> <tr> <td>CAT, MAT, CMAT etc.</td> <td>10%</td> <td>5%</td> </tr> <tr> <td>HPCET-2016</td> <td>20%</td> <td>15%</td> </tr> <tr> <td>Bachelor Degree</td> <td>60%</td> <td>55%</td> </tr> </tbody> </table> <p>Wherein the % indicates the marks obtained by the candidate in the entrance test/Bachelor degree examination out of the total marks assigned to the entrance test/examination.</p>	Name of Exam	General Category	Reserved Categories	CAT, MAT, CMAT etc.	10%	5%	HPCET-2016	20%	15%	Bachelor Degree	60%	55%
Name of Exam	General Category	Reserved Categories											
CAT, MAT, CMAT etc.	10%	5%											
HPCET-2016	20%	15%											
Bachelor Degree	60%	55%											
MCA	<p>1. The admission to MCA first year degree course shall be made on the basis of merit of HPCET-2016.</p> <p>2. In case the seats remain vacant after exhausting the merit of HPCET–2016, the remaining seats will be filled up on the basis of merit of recognized Bachelor Degree of minimum three years duration with Mathematics at 10+2 level or at Graduate level.</p> <p>Note:- The minimum/cut off marks to seek admission to MCA degree course on the basis of HPCET–2016 and Bachelor Degree shall be as follows:</p> <table border="1"> <thead> <tr> <th>Name of Exam</th> <th>General Category</th> <th>Reserved Categories</th> </tr> </thead> <tbody> <tr> <td>HPCET–2016</td> <td>20%</td> <td>15%</td> </tr> <tr> <td>Bachelor Degree</td> <td>60%</td> <td>55%</td> </tr> </tbody> </table> <p>Wherein the % indicates the marks obtained by the candidate in the entrance test/Bachelor Degree examination out of the total marks assigned to the entrance test/examination.</p>	Name of Exam	General Category	Reserved Categories	HPCET–2016	20%	15%	Bachelor Degree	60%	55%			
Name of Exam	General Category	Reserved Categories											
HPCET–2016	20%	15%											
Bachelor Degree	60%	55%											

5. SEATS AVAILABLE

The seats available in different Institutions in undergraduate and post graduate programmes for the year 2015-16 are given in **Appendix-E** which shall normally be followed for session 2016-17.

6. RESERVATION OF SEATS

As per State/Central Government rules, candidates belonging to certain categories are admitted to seats reserved for them based on relaxed criteria. These categories are:

- (a) Other Backward Classes (OBC)- Non Creamy Layer as per the Central list of Other Backward Classes available on National Commission for Backward Classes (NCBC), Government of India website : www.ncbc.nic.in.
- (b) Scheduled Castes (SC)
- (c) Scheduled Tribes (ST)

The benefit of reservation for admission will be given only for Himachal Bonafide applicants and the applicants belonging to other States will be treated under General Category only.

7. SCHEDULE OF ENTRANCE EXAMINATION

HPCET-2016 Entrance Tests for different programmes/courses will be conducted from 14.05.2016 to 15.05.2016 as given in Table (3).

Table (3)

Sr. No.	Course/Programme	Date of Common Entrance Test	Tentative Date of declaration of result
1.	MBA	14.05.2016 (10 AM to 12 Noon)	31.05.2016
2.	MCA	14.05.2016 (2 PM to 4PM)	31.05.2016
3.	B.Tech (Direct Entry) and B. Pharmacy, B. Pharmacy (Ayurveda)	14.05.2016 (10 AM to 1 PM)	31.05.2016
4.	M.Tech & M.Pharmacy	15.05.2016 (2 PM to 5 PM)	31.05.2016

8. EXAMINATION CENTRES

Examination Centres with respective Codes are given below in Table (4)

Table (4)

Examination Centre	Code	Examination Centre	Code	Examination Centre	Code
Bilaspur	01	Nurpur	06	Una	11
Chamba	02	Mandi	07	Palampur	12
Dharamshala	03	Shimla	08	Paonta Sahib	13
Hamirpur	04	Nahan	09	Nalagarh	14
Kullu	05	Solan	10	Chandigarh	15

Note: University reserves the right to add or delete the number of Examination Centres or can create Centes at any other place within/outside the State of Himachal Pradesh keeping in view the number of applicants. The University can change the Examination Centre of the applicant at any time.

9. APPLICATION FORM AND ITS SUBMISSION

9.1 DOWNLOADING OF INFORMATION BROCHURE

All candidates have to apply online. The Information Brochure for HPCET-2016 can be downloaded from the University website at www.himtu.ac.in

9.2 SUBMISSION OF APPLICATION FORM

- (a) The Application Form duly completed should be checked, if all the items have been correctly filled and appropriate codes indicating your exact choice have been put in the right spaces provided against each. Also ensure that you have adhered to the instructions for completing the online Application Form before submitting it.
- (b) Incomplete Application Form shall not be considered and no correspondence shall be made in this regard. The Application Form once submitted cannot be taken back under any circumstances nor shall the Entrance Examination Fee deposited be refunded in any case.
- (c) Any application submitted after the prescribed last date will not be accepted.
- (d) The applicants are required to pay the Entrance Examination Fee given in table at 9.3 through the following mode:-
 - (i) e-Challan mode
 - (ii) Net banking /Credit card/Debit card

9.3 ENTRANCE EXAMINATION FEE

Entrance Examination Fee (**Non-refundable**) for HPCET-2016 is as follows-

Sr. No.	Category	Fee in Rupess
1	SC/ST/BPL	1400/-
2	Others	1550/-

In case the Entrance Examination Fee is paid through e-challan/credit/debit card, the candidate is required to pay an additional processing charges, if any, as per the norms of bank.

10. ADMIT CARD

- (a) The e-Admit card duly signed by Controller of Examination (hereinafter referred as COE) will be made available for candidates on the University website i.e. www.himtu.ac.in. Candidate should enter the Application Form number to download the e- Admit card.
- (b) The e-Admit card will indicate the name of applicant, father's name, photograph of the student, address of the test centre, test date, etc. Discrepancies, if any, must be brought to the notice of the Controller of Examinations, H.P. Technical University, Gandhi Chowk, Hamirpur-177001 (H.P.) immediately.
- (c) Candidates should take a print of the e-Admit card using the print option on A4 size paper only. Please ensure that all information on the e-Admit card including photograph is clearly visible and e-admit card is duly signed by COE.
- (d) Candidates will not be permitted to appear for the test without valid e-Admit card.
- (e) Candidates must not mutilate the e-Admit card or change any entry made therein after it has been authenticated and received by them. Impersonation is a legally punishable offence.
- (f) The e-Admit card is an important document and it must be preserved and produced at the time of test. Candidate should report to the allotted test centre along-with e-Admit card and ID proof, **half an hour before commencement of the examination.**

11. ANSWER KEYS, PROCEDURE TO RESOLVE TIES AND DECLARATION OF RESULT

11.1 ANSWER KEYS

- (a) The answer keys for various entrance examinations shall be available on the University website by 5.00 PM on the respective dates of entrance examinations.
- (b) Candidates can forward their written complaints alongwith supporting documents/solution pertaining to question papers/answer keys and it must reach in the office of the Controller of Examination, H.P. Technical University, Gandhi Chowk, Hamirpur-177001 (H.P.) within two days, from the date of conducting respective examinations, by 5.00 PM through e-mail coehimtu@gmail.com Thereafter, **no complaint in this regard shall be entertained.**

11.2 PROCEDURE TO RESOLVE TIE

Himachal Pradesh Technical University will follow the below mentioned rules to break the ties in ranking procedure if the candidates have scored the same aggregate marks in HPCET 2016 or degree of qualifying examination:

(a) B.Tech/B.Pharma/B.Pharma (Ayurveda)-Direct Entry

- (i) If two applicants have the same HPCET 2016 aggregate marks, the candidate with higher marks in Physics will be ranked above.
- (ii) If the marks in Physics are same, then a higher marks in Chemistry would break the tie.
- (iii) If the marks in Physics and Chemistry are same, then the marks in third subject would be eventually same. In such a case applicant's date of birth will be considered. Elder candidate gets the benefit of being ranked above.

(b) M. Tech

- (i) If two Applicants have the same HPCET 2016 Aggregate Marks, the candidate with higher marks in Aptitude Test will be ranked above.
- (ii) If the Aptitude Test marks are same, then a higher Maths marks would break the tie.
- (iii) If the Aptitude Test and Maths marks are same, then the third subject marks would be eventually same. In such a case applicant's date of birth will be considered. Elder candidate gets the benefit of being ranked above.

(c) M. Pharm.

- (i) If two applicants have the same aggregate marks, the candidate with higher marks in concerned subject i.e. Section B will be ranked above.
- (ii) If the subject marks are same, then the applicant's date of birth will be considered. Elder candidate gets the benefit of being ranked above.

(d) M.B.A & M.C.A - Direct Entry

- (i) If two applicants have the same HPCET-2016 aggregate marks, the candidate with higher marks in Verbal Ability (Section A) will be ranked above.
- (ii) If the Verbal Ability Marks are same, then a higher Quantitative Ability (Section B) marks would break the tie.
- (iii) If the Quantitative Ability marks are same, then a higher Data Interpretation (Section C) marks would break the tie.
- (iv) If the Verbal Ability, Quantitative Ability and Data Interpretation marks are same, then the applicant's date of birth will be considered. Elder candidate gets the benefit of being ranked above.

11.3 RESULT

The result of HPCET-2016 will be uploaded on the University's website www.himtu.ac.in and will also be available on the Notice Board of the University as per the schedule mentioned in Table-3. The result of each candidate will be provisional. The detail of marks can be obtained from the University website.

- (a) The merit shall be determined on the basis of aggregate number of marks/rank obtained in the entrance test by the candidates and the merit list will be prepared accordingly.
- (b) The merit will be declared and displayed on the Notice Board of the University and will also be made available on University website. Merely assigning the marks/rank in the merit list will not confer a right of admission. The admission will be made if the candidate (s) satisfy the prescribed eligibility criteria on verification of the original certificates/documents and availability of seats at the time of counselling/admission.

12. CODE OF CONDUCT FOR CANDIDATES

- (a) Candidates shall maintain perfect silence and attend to their question paper only. Any conversation or gesture or disturbance in the Examination Room/Hall shall be deemed as misbehavior. If a candidate is found using unfair means or impersonating, his/her candidature shall be cancelled and he/she will be liable to be debarred for taking test either permanently or for a specified period according to the nature of offence.
- (b) Candidates are not allowed to carry any textual material, calculators, Docu pen, slide rules, log tables, electronic watches with facilities of calculator, printed or written material, bits of papers, mobile phone, pager or any other device, except the e-Admit Card inside the Examination Room/Hall. If any candidate is in possession of any of the above item, his/her candidature will be treated as unfair means and his/her current test will be cancelled & he/she will also be debarred for future test(s) & the equipment/material will be seized.
- (c) The candidate shall not remove any page(s) from the Test-Booklet (in case of pen and paper based test) and if he/she is found to have removed any page(s) from his/her Test Booklet, he/she will be presumed to have used unfair means and shall be liable for criminal action.

13. GUIDELINES FOR CANDIDATES APPEARING IN HPCET-2016

- (a) Please check the e-Admit Card carefully for your Name, Test Center Name, Place and Category.
- (b) The e-Admit Card is issued provisionally to the candidate subject to his/her satisfying the eligibility conditions.

- (c) The Examination Rooms/Hall will be opened 30 minutes before the commencement of the test. Candidates should take their seats immediately after opening of the Examination Hall. If the candidates do not report in time, they are likely to miss some of the general instructions to be announced in the Examination Hall.
- (d) The candidate must show, on demand, the e-Admit Card and ID Proof for admission in the Examination Room/Hall. A candidate who does not possess the e-Admit Card issued by the University shall not be permitted for the test under any circumstances by the Centre Superintendent.
- (e) For Aptitude Test, candidates are advised to bring their own geometry box set, pencils, erasers etc.
- (f) Candidates are advised to bring with them a cardboard or a clip board on which nothing should be written, so that they have no difficulty in filling responses in the Answer Sheet even if the tables provided in the Examination Room/Hall do have smooth surface. They should also bring with them their own Ball Point Pens (Black/Blue) of good quality.
- (g) No Candidate will be allowed to carry any baggage inside the Examination Centre.
- (h) No candidate, without the special permission of the Centre Superintendent or the Invigilator concerned, will leave his/her seat or Examination Room until the full duration of the paper. Candidates should not leave the Room/Hall without handing over their Answer Sheets to the Invigilators on duty.
- (i) Use of electronic devices like mobile phone, calculator etc. is **NOT PERMITTED** in HPCET-2016. Materials like log table, book, notebook, etc. should NOT be brought into the Examination Hall/Room.

14. PROCEDURE TO FILL UP ONLINE APPLICATION FORM FOR HPCET-2016

14.1 READ THE INSTRUCTIONS CAREFULLY BEFORE PROCEEDING AHEAD FOR REGISTRATION

- (a) Fresh Registration has been started *w.e.f.* 2nd March 2016 through HPCET - 2016
- (b) Last date for receiving the scanned copy of the Application Form duly signed by their parents through e-mail on himtuadmission@gmail.com on or before 2nd May, 2016.
- (c) The candidate is required to pay prescribed Entrance Examination Fee.
- (d) For better accuracy, please use only Firefox, Chrome & Internet Explorer.
- (e) Clear the temporary internet files and cookies before filling up each form.

- (f) Do not fill the Application Form using mobile phones.
- (g) Fill up the Application Form at a suitable time and from a location where you have a good bandwidth available.
- (h) Do not open more than one session at the same time on the same computer while filling the online application form. *i.e*; do not fill more than one form at the same time on the same computer even if you are filling the two applications using different browsers or different tabs of the same browser.
- (i) Candidates can submit **ONLINE** application form to seek admission in a particular course through the concerned link available on the University website <http://www.himtu.ac.in>.
- (j) Fill in the application form only if you meet the eligibility and admission criteria for the program.
- (k) The candidates are advised to read the instructions carefully before submitting the application form. The instructions are self-explanatory and follow them strictly.
- (l) You are required to have a scanned copy of your coloured photograph (3.5 x 4.5 cm) and digital signature. The scanned copies of coloured photo (3.5 x 4.5 cm) in (JPEG /JPG) Format and signature (in *.JPEG /*.JPG) Format are required to be uploaded during the **ONLINE** submission of application form. The file size of photograph and signature in any case shall not exceed 50kb and 30kb for each file separately.
- (m) It is mandatory to provide your cell number and email address as it will be used for registration. Mobile No. and e-mail address are required for making any communication related to your admission by the Admission Office.
- (n) Incomplete online application form and printout (PDF) received after the due date will be rejected straightway. No correspondence in this regard shall be entertained.
- (o) Himachal Pradesh Technical University does not take any responsibility for delay or loss of material during post/transit.

14.2 PROCEDURE FOR SUBMISSION OF ONLINE APPLICATION FORM

STEP 1: OPEN ACCOUNT

- Open the web-site <http://www.himtu.ac.in> and Click the button “**Apply Online**”. Ensure that you have read all the instructions contained in the Information Brochure HPCET-2016 and instruction displaying on the webpage carefully.
- Click “**Registered Now**” icon on the left of the screen. Account Registration Form will appear. Enter the essential information against User Name (The Username must be unique). The candidates are advised to use some special character during making entry in the Username, Password, Confirm password, Date of Birth, email address and mobile number and Click on the Create Account. The system will generate a Unique Form Number which will be displayed on the screen. The message of Username and Form No generated by the system will also be delivered on e-mail address provided during the registration. The candidates are advised to note down the Form Number, User Name, Password & Date of Birth for future reference.

STEP 2: LOGIN TO FILL THE APPLICATION FORM

Click the “**login**” button. Fill the username, password and Application Form Number and Date of Birth. Online Application Form will appear. The Form has been divided into four sections i.e Basic Form Detail, Upload Photo and Signature, Fee & Fee payment Detail and Confirmation Page. After successfully submission of Basic Form detail, next page will open automatically.

BASIC FORM DETAIL:

- **NAME OF THE CANDIDATE:** Enter your name in capital letters as it appears in your Matric or first Board/University Examination Certificate.
- **FATHER’S NAME:** Write your father’s name in capital letters as it appears in your matric or first Board/University Examination Certificate.
- **MOTHER’S NAME:** Write your mother’s name in capital letters as it appears in your matric or first Board/University Examination Certificate.
- **NATIONALITY:** Select the appropriate option *Indian/Other*, whichever is applicable.
- **SEX:** Select the appropriate option - *Male/Female/Other*, whichever is applicable.

- **PROGRAMME APPLIED FOR :** Select the appropriate program from the dropdown for which you want to apply-B.Tech(only)/B.Pharma Allo.(only)/B.Pharma Ayur.(only)/B.Tech & B.Pharma Allo(Both)/B.Pharma Ayur. & B.Pharma Allo (Both)/M.Pharmacy/MBA/MCA/M.Tech. The candidates who are eligible and want to apply to appear in two tests simultaneously must select B.Tech & B.Pharma Allo(Both) / B.Pharma Ayurveda & B.Pharma Allopathy (Both) from the dropdown menu and have to pay double Entrance Examination Fee.
- **BRANCH :** On selection of Programme its related applicable branches, if any, will automatically appear. Select the applicable branch from the dropdown menu.
- **BETI HAI ANMOL:** Select *Yes/No* option whichever is applicable.
- **KASHMIRI MIGRANTS:** Select the option *Yes/No* whichever is applicable
- **QUOTA UNDER WHICH APPLIED:** Select the option *All India Quota/ HP State Quota* from the dropdown whichever is applicable.
- **EXAMINATION CENTRE:** Select the convenient Examination Centre from the dropdown menu.
- **BONAFIED/DOMICILE OF HIMACHAL:** Select *Yes/No* option, whichever is applicable.
- **CATEGORY:** Select category *GEN/SC/ST/OBC*, whichever is applicable.
- **SUB CATEGORY:** Tick on the appropriate checkboxes displaying in front of IRDP/BPL/Antodaya/Orphan, Physically Challenged, Sports & Freedom Fighter if applicable.
- **DATE OF BIRTH, EMAIL ID AND MOBILE NUMBER:** The data in these fields will automatically appear as you have filled it during the registration of this form.
- **DEFENCE, WEIGHTAGE CODE FOR DEFENCE:** Select Yes from the dropdown if you belong to the wards of defence personnel, The “**Weightage code for Defence**” field will be enabled on selection of “Yes” from the defence field. You have to select the appropriate option from the dropdown showing the contents: Deceased in war/action, Disabled during war/action, Death attributable to military service, Disabled during service (Medically Board out due to disability attributable to military service), Gallantry awardee during war/action/service, Ex-servicemen or personnel in service.

- **BACKWARD AREA, WEIGHTAGE CODE FOR BACKWARD AREA:** Select Yes from the dropdown if you belong to the backward area. “**Weightage code for Backward Area**” field will be enabled on selection of “Yes” from the this field. You have to select the appropriate option from the dropdown which will show the contents : Passed at least two examinations from schools located in the backward area, Passed at least one examination from school located in the backward area, Passed such two examinations from a schools located in the immediately adjoining backward area, Passed such one examination from a school located in the immediately adjoining backward area, Not passed any examination from backward area but belongs to backward area.
- **TUITION FEE WAIVER SCHEME, IF YES THEN IN WHICH CATEGORY:** Select the appropriate option. On selection of yes, the field “**If Yes then in which category**” will become enabled and you have to select any one option from the dropdown displaying the contents : IRDP, BPL, Antodaya, Orphan & Other.
- **QUALIFYING EXAM. OR IT’S EQUIVALENT:** Select the Qualifying Exam or its equivalent-10+2(Medical)/10+2(NM)/B.Tech/B. Pharma/BA/B.Sc/B.Com/BBA/BCA/B.Sc.(IT/Computer Science)/Diploma in Engg./Diploma in Pharmacy (Allopathy)/any other-whichever is applicable. Then select the Passed/Appeared option and enter the relevant marks detail. On selection of undergraduate courses from the program opted and on selection of passed from the qualifying Exam or its equivalent field a table for entering subject wise marks detail will automatically appear. Fill the contents in the Table.
- **PERCENTAGE IN QUALIFYING EXAMINATION OR ITS EQUIVALENT:** Fill the percentage of marks obtained in the qualifying examination.
- **PERMANENT ADDRESS:** Enter your permanent address. The address must include your name, father’s name, village, post office, Tehsil, district including the PIN Code.
- **CORRESPONDENCE ADDRESS:** Enter your postal address for communication. The address must include your name, c/o name if required, etc. including the PIN Code.
- Click “**Save & Continue**”. Thereafter the option **UPLOAD PHOTOGRAHP/ SIGNATURE** will appear on the screen.

STEP 3: UPLOAD PHOTOGRAPH AND SIGNATURE

- **PHOTOGRAPH:** Click on **browse** to select your latest scanned photograph and select the file size upto 50kb of *.JPEG/*.JPG format from the source.
- **SIGNATURE :** Upload the JPEG/JPG format file by browsing the source and select your latest digital signatures of file size upto 30kb.

After attaching the files, click on “**Upload**” button. You will see the webpage message of **Photograph and Signature uploaded Successfully** on the screen. The next page **Fee payment** will appear automatically.

STEP 4: MAKE FEE PAYMENT

- Click the “Click here to get Fee Payment Link” Button; two option will appear to make fee payment. i.e. “Offline link to generate PNB Challan” and “Online Fee Payment Link” will appear. If you want to make payment through bank Challan, select “Offline link to generate PNB Challan” option. PNB challan in triplicate (i.e. candidate copy, bank copy and university copy) will be generated. Take the print out of the challan and deposit the Fee in the nearest PNB branch. After depositing the Fee in the bank, re-login in your account at least after 24 hours; and enter transaction number provided by the bank on the PNB Challan in the payment details page and click on “Save Payment Details”. The form will proceed on confirmation page.
- If you are interested to make fee payment online mode, Click the button “Click here to Fee payment”. Before proceeding further read the “Terms & Conditions” carefully and choose a payment method from HDFC Credit / Debit Card, Other Credit /Other Debit Card, Net banking then select Debit / Credit Type i.e. Visa Debit Cards (All Banks), MasterCard Debit Cards (All Banks), Other Maestro Card and make necessary entries of Card Number, Name of the Card, CVV Number & expiry date etc. click on “Pay Now” button. You will be automatically redirected to your bank website. After successful transaction the web page will automatically redirected to HimTU university website <http://www.himtu.ac.in> If you opt the fee payment method “Net Banking” you have to select the name of the bank from the dropdown menu and have to enter your bank Login Id and password, make the necessary entries in the payment form of the concerned bank. Click on “Pay Now” button. You will be automatically redirected to your bank website. After successful transaction the web page will redirected to HimTU university website.
- In case your Registration Processing Fee has been debited from your account and you are not able to get the PDF, please pay again to get the PDF. However, any multiple payments received by us for the same student (applying more than once with the same Registration Number) will be refunded by the office on its own at the end of the online forms process. For

refund contact at email finofficerhimtu@gmail.com instead of customer care of the concerned bank.

STEP 5: CONFIRMATION

Thereafter, filled Application Form under the “**Confirmation**” button will appear. At the bottom of the Application Form two options i.e. “**Edit**” and “**Final Submit**” shall be there. In case you intend to make correction on the confirmation page, click on “**Edit**” button. The data filled by you will appear in editable form. Make the necessary correction and click on ‘**Update Details**’ button to save the changes. Now click the “**Final Submit**” button. It may be ensured before you click the “**Final Submit**” button that all the applicable fields have been filled correctly.

- Thereafter, screen will display the full details entered by you in the Application Form along with declaration. Take a print out of the Application Form(PDF) and get it signed from your parents/guardian and send it duly scanned through e-mail ID himtuadmission@gmail.com

IMPORTANT NOTE:

- Do not make any changes with the pen/pencil on the printed Application Form, otherwise your Application Form will be rejected straight forward.
- Before sending the form to Himachal Pradesh Technical University, the applicant should make sure that the original Application Form is duly countersigned by the parent/guardian on the space provided.

HPCET-2016 Syllabus for B.Tech./B.Pharmacy/B. Pharmacy (Ayurveda)**MATHEMATICS****UNIT 1: SETS, RELATIONS AND FUNCTIONS:**

Sets and their representation; Union, intersection and complement of sets and their algebraic properties; Power set; Relation, Types of relations, equivalence relations, functions; one-one, into and onto functions, composition of functions.

UNIT 2: COMPLEX NUMBERS AND QUADRATIC EQUATIONS:

Complex numbers as ordered pairs of reals, Representation of complex numbers in the form $a+ib$ and their representation in a plane, Argand diagram, algebra of complex numbers, modulus and argument (or amplitude) of a complex number, square root of a complex number, triangle inequality, Quadratic equations in real and complex number system and their solutions. Relation between roots and co-efficients, nature of roots, formation of quadratic equations with given roots.

UNIT 3: MATRICES AND DETERMINANTS:

Matrices, algebra of matrices, types of matrices, determinants and matrices of order two and three. Properties of determinants, evaluation of determinants, area of triangles using determinants. Adjoint and evaluation of inverse of a square matrix using determinants and elementary transformations, Test of consistency and solution of simultaneous linear equations in two or three variables using determinants and matrices.

UNIT 4: PERMUTATIONS AND COMBINATIONS:

Fundamental principle of counting, permutation as an arrangement and combination as selection, Meaning of $P(n,r)$ and $C(n,r)$, simple applications.

UNIT 5: MATHEMATICAL INDUCTION:

Principle of Mathematical Induction and its simple applications.

UNIT 6: BINOMIAL THEOREM AND ITS SIMPLE APPLICATIONS:

Binomial theorem for a positive integral index, general term and middle term, properties of Binomial coefficients and simple applications.

UNIT 7: SEQUENCES AND SERIES:

Arithmetic and Geometric progressions, insertion of arithmetic, geometric means between two given numbers. Relation between A.M. and G.M. Sum upto n terms of special series: S_n , S_{n^2} , S_{n^3} . Arithmetic - Geometric progression.

UNIT 8: LIMIT, CONTINUITY AND DIFFERENTIABILITY:

Real - valued functions, algebra of functions, polynomials, rational, trigonometric, logarithmic and exponential functions, inverse functions. Graphs of simple functions. Limits, continuity and differentiability. Differentiation of the sum, difference, product and quotient of two functions. Differentiation of trigonometric, inverse trigonometric, logarithmic, exponential, composite and implicit functions; derivatives of order upto two. Rolle's and Lagrange's Mean Value Theorems. Applications of derivatives: Rate of change of quantities, monotonic - increasing and decreasing functions, Maxima and minima of functions of one variable, tangents and normals.

UNIT 9: INTEGRAL CALCULUS:

Integral as an anti - derivative. Fundamental integrals involving algebraic, trigonometric, exponential and logarithmic functions. Integration by substitution, by parts and by partial fractions. Integration using trigonometric identities.

Evaluation of simple integrals of the type

$$\int \frac{dx}{x^2 \pm a^2}, \int \frac{dx}{\sqrt{x^2 \pm a^2}}, \int \frac{dx}{a^2 - x^2}, \int \frac{dx}{\sqrt{a^2 - x^2}}, \int \frac{dx}{ax^2 + bx + c},$$

$$\int \frac{dx}{\sqrt{ax^2 + bx + c}}, \int \frac{(px+q)dx}{ax^2 + bx + c}, \int \frac{(px+q)dx}{\sqrt{ax^2 + bx + c}}$$

$$\int \sqrt{a^2 \pm x^2} dx \quad \int \sqrt{x^2 - a^2} dx$$

Integral as limit of a sum. Fundamental Theorem of Calculus. Properties of definite integrals. Evaluation of definite integrals, determining areas of the regions bounded by simple curves in standard form.

UNIT 10: DIFFERENTIAL EQUATIONS:

Ordinary differential equations, their order and degree. Formation of differential equations. Solution of differential equations by the method of separation of variables, solution of homogeneous and linear differential equations of the type:

$$\frac{dy}{dx} + p(x)y = q(x)$$

UNIT 11: CO-ORDINATE GEOMETRY:

Cartesian system of rectangular co-ordinates in a plane, distance formula, section formula, locus and its equation, translation of axes, slope of a line, parallel and perpendicular lines, intercepts of a line on the coordinate axes.

Straight lines

Various forms of equations of a line, intersection of lines, angles between two lines, conditions for concurrence of three lines, distance of a point from a line, equations of internal and external bisectors of angles between two lines, coordinates of centroid, orthocentre and circumcentre of a triangle, equation of family of lines passing through the point of intersection of two lines.

Circles, conic sections

Standard form of equation of a circle, general form of the equation of a circle, its radius and centre, equation of a circle when the end points of a diameter are given, points of intersection of a line and a circle with the centre at the origin and condition for a line to be tangent to a circle, equation of the tangent. Sections of cones, equations of conic sections (parabola, ellipse and hyperbola) in standard forms, condition for $y = mx + c$ to be a tangent and point (s) of tangency.

UNIT 12: THREE DIMENSIONAL GEOMETRY:

Coordinates of a point in space, distance between two points, section formula, direction ratios and direction cosines, angle between two intersecting lines. Skew lines, the shortest distance between them and its equation. Equations of a line and a plane in different forms, intersection of a line and a plane, coplanar lines.

UNIT 13: VECTOR ALGEBRA:

Vectors and scalars, addition of vectors, components of a vector in two dimensions and three dimensional space, scalar and vector products, scalar and vector triple product.

UNIT 14: STATISTICS AND PROBABILITY:

Measures of Dispersion: Calculation of mean, median, mode of grouped and ungrouped data calculation of standard deviation, variance and mean deviation for grouped and ungrouped data.

Probability: Probability of an event, addition and multiplication theorems of probability, Baye's theorem, probability distribution of a random variate, Bernoulli trials and Binomial distribution.

UNIT 15: TRIGONOMETRY:

Trigonometrical identities and equations. Trigonometrical functions. Inverse trigonometrical functions and their properties. Heights and Distances.

UNIT 16: MATHEMATICAL REASONING:

Statements, logical operations and, or, implies, implied by, if and only if. Understanding of tautology, contradiction, converse and contrapositive.

PHYSICS

The syllabus contains two Sections - A and B. Section - A pertains to the Theory Part having 80% weightage, while Section - B contains Practical Component (Experimental Skills) having 20% weightage.

SECTION - A

UNIT 1: PHYSICS AND MEASUREMENT

Physics, technology and society, S I units, Fundamental and derived units. Least count, accuracy and precision of measuring instruments, Errors in measurement, Dimensions of Physical quantities, dimensional analysis and its applications.

UNIT 2: KINEMATICS

Frame of reference. Motion in a straight line: Position-time graph, speed and velocity. Uniform and non-uniform motion, average speed and instantaneous velocity Uniformly accelerated motion, velocity-time, position-time graphs, relations for uniformly accelerated motion. Scalars and Vectors, Vector addition and Subtraction, Zero Vector, Scalar and Vector products, Unit Vector, Resolution of a Vector. Relative Velocity, Motion in a plane, Projectile Motion, Uniform Circular Motion.

UNIT 3: LAWS OF MOTION

Force and Inertia, Newton's First Law of motion; Momentum, Newton's Second Law of motion; Impulse; Newton's Third Law of motion. Law of conservation of linear momentum and its applications, Equilibrium of concurrent forces.

Static and Kinetic friction, laws of friction, rolling friction.

Dynamics of uniform circular motion: Centripetal force and its applications.

UNIT 4: WORK, ENERGY AND POWER

Work done by a constant force and a variable force; kinetic and potential energies, work energy theorem, power.

Potential energy of a spring, conservation of mechanical energy, conservative and nonconservative forces; Elastic and inelastic collisions in one and two dimensions.

UNIT 5: ROTATIONAL MOTION

Centre of mass of a two-particle system, Centre of mass of a rigid body; Basic concepts of rotational motion; moment of a force, torque, angular momentum, conservation of angular momentum and its applications; moment of inertia, radius of gyration. Values of moments of inertia for simple geometrical

objects, parallel and perpendicular axes theorems and their applications. Rigid body rotation, equations of rotational motion.

UNIT 6: GRAVITATION

The universal law of gravitation. Acceleration due to gravity and its variation with altitude and depth. Kepler's laws of planetary motion. Gravitational potential energy; gravitational potential. Escape velocity. Orbital velocity of a satellite. Geo-stationary satellites.

UNIT 7: PROPERTIES OF SOLIDS AND LIQUIDS

Elastic behaviour, Stress-strain relationship, Hooke's Law, Young's modulus, bulk modulus, modulus of rigidity. Pressure due to a fluid column; Pascal's law and its applications. Viscosity, Stokes' law, terminal velocity, streamline and turbulent flow, Reynolds number. Bernoulli's principle and its applications. Surface energy and surface tension, angle of contact, application of surface tension - drops, bubbles and capillary rise. Heat, temperature, thermal expansion; specific heat capacity, calorimetry; change of state, latent heat. Heat transfer-conduction, convection and radiation, Newton's law of cooling.

UNIT 8: THERMODYNAMICS

Thermal equilibrium, zeroth law of thermodynamics, concept of temperature. Heat, work and internal energy. First law of thermodynamics. Second law of thermodynamics: reversible and irreversible processes. Carnot engine and its efficiency.

UNIT 9: KINETIC THEORY OF GASES

Equation of state of a perfect gas, work done on compressing a gas. Kinetic theory of gases - assumptions, concept of pressure. Kinetic energy and temperature: rms speed of gas molecules; Degrees of freedom, Law of equipartition of energy, applications to specific heat capacities of gases; Mean free path, Avogadro's number.

UNIT 10: OSCILLATIONS AND WAVES

Periodic motion - period, frequency, displacement as a function of time. Periodic functions. Simple harmonic motion (S.H.M.) and its equation; phase; oscillations of a spring - restoring force and force constant; energy in S.H.M. - kinetic and potential energies; Simple pendulum - derivation of expression for its time period; Free, forced and damped oscillations, resonance.

Wave motion. Longitudinal and transverse waves, speed of a wave. Displacement relation for a progressive wave. Principle of superposition of waves, reflection of waves, Standing waves in strings and organ pipes, fundamental mode and harmonics, Beats, Doppler effect in sound

UNIT 11: ELECTROSTATICS

Electric charges: Conservation of charge, Coulomb's law-forces between two point charges, forces between multiple charges; superposition principle and continuous charge distribution.

Electric field: Electric field due to a point charge, Electric field lines, Electric dipole, Electric field due to a dipole, Torque on a dipole in a uniform electric field.

Electric flux, Gauss's law and its applications to find field due to infinitely long uniformly charged straight wire, uniformly charged infinite plane sheet and uniformly charged thin spherical shell. Electric potential and its calculation for a point charge, electric dipole and system of charges; Equipotential surfaces, Electrical potential energy of a system of two point charges in an electrostatic field.

Conductors and insulators, Dielectrics and electric polarization, capacitor, combination of capacitors in series and in parallel, capacitance of a parallel plate capacitor with and without dielectric medium between the plates, Energy stored in a capacitor.

UNIT 12: CURRENT ELECTRICITY

Electric current, Drift velocity, Ohm's law, Electrical resistance, Resistances of different materials, V-I characteristics of Ohmic and nonohmic conductors, Electrical energy and power, Electrical resistivity, Colour code for resistors; Series and parallel combinations of resistors; Temperature dependence of resistance.

Electric Cell and its Internal resistance, potential difference and emf of a cell, combination of cells in series and in parallel. Kirchhoff's laws and their applications. Wheatstone bridge, Metre bridge. Potentiometer - principle and its applications.

UNIT 13: MAGNETIC EFFECTS OF CURRENT AND MAGNETISM

Biot - Savart law and its application to current carrying circular loop. Ampere's law and its applications to infinitely long current carrying straight wire and solenoid. Force on a moving charge in uniform magnetic and electric fields. Cyclotron.

Force on a current-carrying conductor in a uniform magnetic field. Force between two parallel current-carrying conductors-definition of ampere. Torque experienced by a current loop in uniform magnetic field; Moving coil galvanometer, its current sensitivity and conversion to ammeter and voltmeter.

Current loop as a magnetic dipole and its magnetic dipole moment. Bar magnet as an equivalent solenoid, magnetic field lines; Earth's magnetic field and magnetic elements. Para-, dia- and ferro- magnetic substances.

Magnetic susceptibility and permeability, Hysteresis, Electromagnets and permanent magnets.

UNIT 14: ELECTROMAGNETIC INDUCTION AND ALTERNATING CURRENTS

Electromagnetic induction; Faraday's law, induced emf and current; Lenz's Law, Eddy currents. Self and mutual inductance. Alternating currents, peak and rms value of alternating current/ voltage; reactance and impedance; LCR series circuit, resonance; Quality factor, power in AC circuits, wattless current. AC generator and transformer.

UNIT 15: ELECTROMAGNETIC WAVES

Electromagnetic waves and their characteristics. Transverse nature of electromagnetic waves.

Electromagnetic spectrum (radio waves, microwaves, infrared, visible, ultraviolet, Xrays, gamma rays). Applications of e.m. waves.

UNIT 16: OPTICS

Reflection and refraction of light at plane and spherical surfaces, mirror formula, Total internal reflection and its applications, Deviation and Dispersion of light by a prism, Lens Formula, Magnification, Power of a Lens, Combination of thin lenses in contact, Microscope and Astronomical Telescope (reflecting and refracting) and their magnifying powers.

Wave optics: wavefront and Huygens' principle, Laws of reflection and refraction using Huygen's principle. Interference, Young's double slit experiment and expression for fringe width, coherent sources and sustained interference of light. Diffraction due to a single slit, width of central maximum. Resolving power of microscopes and astronomical telescopes, Polarisation, plane polarized light; Brewster's law, uses of plane polarized light and Polaroids.

UNIT 17: DUAL NATURE OF MATTER AND RADIATION

Dual nature of radiation. Photoelectric effect, Hertz and Lenard's observations; Einstein's photoelectric equation; particle nature of light. Matter waves-wave nature of particle, de Broglie relation. Davisson-Germer experiment.

UNIT 18: ATOMS AND NUCLEI

Alpha-particle scattering experiment; Rutherford's model of atom; Bohr model, energy levels, hydrogen spectrum. Composition and size of nucleus, atomic masses, isotopes, isobars; isotones. Radioactivity-alpha, beta and gamma particles/rays and their properties; radioactive decay law. Mass-energy relation, mass defect; binding energy per nucleon and its variation with mass number, nuclear fission and fusion.

UNIT 19: ELECTRONIC DEVICES

Semiconductors; semiconductor diode: I-V characteristics in forward and reverse bias; diode as a rectifier; I-V characteristics of LED, photodiode, solar cell and Zener diode; Zener diode as a voltage regulator. Junction transistor, transistor action, characteristics of a transistor; transistor as an amplifier (common emitter configuration) and oscillator. Logic gates (OR, AND, NOT, NAND and NOR). Transistor as a switch.

UNIT 20: COMMUNICATION SYSTEMS

Propagation of electromagnetic waves in the atmosphere; Sky and space wave propagation, Need for modulation, Amplitude and Frequency Modulation, Bandwidth of signals, Bandwidth of Transmission medium, Basic Elements of a Communication System (Block Diagram only).

SECTION-B

UNIT 21: EXPERIMENTAL SKILLS

Familiarity with the basic approach and observations of the experiments and activities:

1. Vernier callipers-its use to measure internal and external diameter and depth of a vessel.
2. Screw gauge-its use to determine thickness/ diameter of thin sheet/wire.
3. Simple Pendulum-dissipation of energy by plotting a graph between square of amplitude and time.
4. Metre Scale - mass of a given object by principle of moments.
5. Young's modulus of elasticity of the material of a metallic wire.
6. Surface tension of water by capillary rise and effect of detergents.
7. Co-efficient of Viscosity of a given viscous liquid by measuring terminal velocity of a given spherical body.
8. Plotting a cooling curve for the relationship between the temperature of a hot body and time.
9. Speed of sound in air at room temperature using a resonance tube.
10. Specific heat capacity of a given (i) solid and (ii) liquid by method of mixtures.
11. Resistivity of the material of a given wire using metre bridge.
12. Resistance of a given wire using Ohm's law.
13. Potentiometer -
 - (i) Comparison of emf of two primary cells.
 - (ii) Determination of internal resistance of a cell.

14. Resistance and figure of merit of a galvanometer by half deflection method.
15. Focal length of:
 - (i) Convex mirror
 - (ii) Concave mirror, and
 - (iii) Convex lens
 using parallax method.
16. Plot of angle of deviation vs angle of incidence for a triangular prism.
17. Refractive index of a glass slab using a travelling microscope.
18. Characteristic curves of a p-n junction diode in forward and reverse bias.
19. Characteristic curves of a Zener diode and finding reverse break down voltage.
20. Characteristic curves of a transistor and finding current gain and voltage gain.
21. Identification of Diode, LED, Transistor, IC, Resistor, Capacitor from mixed collection of such items.
22. Using multimeter to:
 - (i) Identify base of a transistor
 - (ii) Distinguish between npn and pnp type transistor
 - (iii) See the unidirectional flow of current in case of a diode and an LED.
 - (iv) Check the correctness or otherwise of a given electronic component (diode, transistor or IC).

CHEMISTRY

SECTION: A

PHYSICAL CHEMISTRY

UNIT 1: SOME BASIC CONCEPTS IN CHEMISTRY

Matter and its nature, Dalton's atomic theory; Concept of atom, molecule, element and compound; Physical quantities and their measurements in Chemistry, precision and accuracy, significant figures, S.I. Units, dimensional analysis; Laws of chemical combination; Atomic and molecular masses, mole concept, molar mass, percentage composition, empirical and molecular formulae; Chemical equations and stoichiometry.

UNIT 2: STATES OF MATTER

Classification of matter into solid, liquid and gaseous states.

Gaseous State:

Measurable properties of gases; Gas laws - Boyle's law, Charles's law, Graham's law of diffusion, Avogadro's law, Dalton's law of partial pressure; Concept of

Absolute scale of temperature; Ideal gas equation; Kinetic theory of gases (only postulates); Concept of average, root mean square and most probable velocities; Real gases, deviation from Ideal behaviour, compressibility factor and van der Waals equation.

Liquid State:

Properties of liquids - vapour pressure, viscosity and surface tension and effect of temperature on them (qualitative treatment only).

Solid State:

Classification of solids: molecular, ionic, covalent and metallic solids, amorphous and crystalline solids (elementary idea); Bragg's Law and its applications; Unit cell and lattices, packing in solids (fcc, bcc and hcp lattices), voids, calculations involving unit cell parameters, imperfection in solids; Electrical and magnetic properties.

UNIT 3: ATOMIC STRUCTURE

Thomson and Rutherford atomic models and their limitations; Nature of electromagnetic radiation, photoelectric effect; Spectrum of hydrogen atom, Bohr model of hydrogen atom - its postulates, derivation of the relations for energy of the electron and radii of the different orbits, limitations of Bohr's model; Dual nature of matter, de-Broglie's relationship, Heisenberg uncertainty principle. Elementary ideas of quantum mechanics, quantum mechanical model of atom, its important features. Concept of atomic orbitals as one electron wave functions; Variation of ψ and ψ^2 with r for 1s and 2s orbitals; various quantum numbers (principal, angular momentum and magnetic quantum numbers) and their significance; shapes of s, p and d - orbitals, electron spin and spin quantum number; Rules for filling electrons in orbitals - aufbau principle, Pauli's exclusion principle and Hund's rule, electronic configuration of elements, extra stability of half-filled and completely filled orbitals.

UNIT 4: CHEMICAL BONDING AND MOLECULAR STRUCTURE

Kossel - Lewis approach to chemical bond formation, concept of ionic and covalent bonds.

Ionic Bonding: Formation of ionic bonds, factors affecting the formation of ionic bonds; calculation of lattice enthalpy.

Covalent Bonding: Concept of electronegativity, Fajan's rule, dipole moment; Valence Shell Electron Pair Repulsion (VSEPR) theory and shapes of simple molecules.

Quantum mechanical approach to covalent bonding: Valence bond theory - Its important features, concept

of hybridization involving s, p and d orbitals; Resonance.

Molecular Orbital Theory - Its important features, LCAOs, types of molecular orbitals (bonding, antibonding), sigma and pi-bonds, molecular orbital electronic configurations of homonuclear diatomic molecules, concept of bond order, bond length and bond energy.

Elementary idea of metallic bonding. Hydrogen bonding and its applications.

UNIT 5: CHEMICAL THERMODYNAMICS

Fundamentals of thermodynamics: System and surroundings, extensive and intensive properties, state functions, types of processes.

First law of thermodynamics - Concept of work, heat internal energy and enthalpy, heat capacity, molar heat capacity; Hess's law of constant heat summation; Enthalpies of bond dissociation, combustion, formation, atomization, sublimation, phase transition, hydration, ionization and solution.

Second law of thermodynamics; Spontaneity of processes; ΔS of the universe and ΔG of the system as criteria for spontaneity, ΔG° (Standard Gibbs energy change) and equilibrium constant.

UNIT 6: SOLUTIONS

Different methods for expressing concentration of solution - molality, molarity, mole fraction, percentage (by volume and mass both), vapour pressure of solutions and Raoult's Law - Ideal and non-ideal solutions, vapour pressure - composition, plots for ideal and non-ideal solutions; Colligative properties of dilute solutions - relative lowering of vapour pressure, depression of freezing point, elevation of boiling point and osmotic pressure; Determination of molecular mass using colligative properties; Abnormal value of molar mass, van't Hoff factor and its significance.

UNIT 7: EQUILIBRIUM

Meaning of equilibrium, concept of dynamic equilibrium.

Equilibria involving physical processes: Solid-liquid, liquid - gas and solid - gas equilibria, Henry's law, general characteristics of equilibrium involving physical processes.

Equilibria involving chemical processes: Law of chemical equilibrium, equilibrium constants (K_p and K_c) and their significance, significance of ΔG and ΔG° in chemical equilibria, factors affecting equilibrium concentration, pressure, temperature, effect of catalyst; Le Chatelier's principle.

Ionic equilibrium: Weak and strong electrolytes, ionization of electrolytes, various concepts of acids and bases (Arrhenius, Brønsted - Lowry and Lewis) and their ionization, acid - base equilibria (including multistage ionization) and ionization constants, ionization of water, pH scale, common ion effect, hydrolysis of salts and pH of their solutions, solubility of sparingly soluble salts and solubility products, buffer solutions.

UNIT 8: REDOX REACTIONS AND ELECTROCHEMISTRY

Electronic concepts of oxidation and reduction, redox reactions, oxidation number, rules for assigning oxidation number, balancing of redox reactions.

Electrolytic and metallic conduction, conductance in electrolytic solutions, molar conductivities and their variation with concentration: Kohlrausch's law and its applications.

Electrochemical cells - Electrolytic and Galvanic cells, different types of electrodes, electrode potentials including standard electrode potential, half - cell and cell reactions, emf of a Galvanic cell and its measurement; Nernst equation and its applications; Relationship between cell potential and Gibbs' energy change; Dry cell and lead accumulator; Fuel cells.

UNIT 9: CHEMICAL KINETICS

Rate of a chemical reaction, factors affecting the rate of reactions: concentration, temperature, pressure and catalyst; elementary and complex reactions, order and molecularity of reactions, rate law, rate constant and its units, differential and integral forms of zero and first order reactions, their characteristics and half - lives, effect of temperature on rate of reactions - Arrhenius theory, activation energy and its calculation, collision theory of bimolecular gaseous reactions (no derivation).

UNIT 10: SURFACE CHEMISTRY

Adsorption- Physisorption and chemisorption and their characteristics, factors affecting adsorption of gases on solids - Freundlich and Langmuir adsorption isotherms, adsorption from solutions.

Catalysis - Homogeneous and heterogeneous, activity and selectivity of solid catalysts, enzyme catalysis and its mechanism.

Colloidal state- distinction among true solutions, colloids and suspensions, classification of colloids - lyophilic, lyophobic; multimolecular, macromolecular and associated colloids (micelles), preparation and properties of colloids - Tyndall effect, Brownian movement, electrophoresis, dialysis, coagulation and flocculation; Emulsions and their characteristics.

SECTION - B

INORGANIC CHEMISTRY

UNIT 11: CLASSIFICATION OF ELEMENTS AND PERIODICITY IN PROPERTIES

Modern periodic law and present form of the periodic table, s, p, d and f block elements, periodic trends in properties of elements atomic and ionic radii, ionization enthalpy, electron gain enthalpy, valence, oxidation states and chemical reactivity.

UNIT 12: GENERAL PRINCIPLES AND PROCESSES OF ISOLATION OF METALS

Modes of occurrence of elements in nature, minerals, ores; Steps involved in the extraction of metals - concentration, reduction (chemical and electrolytic methods) and refining with special reference to the extraction of Al, Cu, Zn and Fe; Thermodynamic and electrochemical principles involved in the extraction of metals.

UNIT 13: HYDROGEN

Position of hydrogen in periodic table, isotopes, preparation, properties and uses of hydrogen; Physical and chemical properties of water and heavy water; Structure, preparation, reactions and uses of hydrogen peroxide; Classification of hydrides - ionic, covalent and interstitial; Hydrogen as a fuel.

UNIT 14: S - BLOCK ELEMENTS (ALKALI AND ALKALINE EARTH METALS)

Group - 1 and 2 Elements

General introduction, electronic configuration and general trends in physical and chemical properties of elements, anomalous properties of the first element of each group, diagonal relationships.

Preparation and properties of some important compounds - sodium carbonate and sodium hydroxide and sodium hydrogen carbonate; Industrial uses of lime, limestone, Plaster of Paris and cement; Biological significance of Na, K, Mg and Ca.

UNIT 15: P - BLOCK ELEMENTS

Group - 13 to Group 18 Elements

General Introduction: Electronic configuration and general trends in physical and chemical properties of elements across the periods and down the groups; unique behaviour of the first element in each group.

Groupwise study of the p - block elements

Group - 13

Preparation, properties and uses of boron and aluminium; Structure, properties and uses of borax, boric acid, diborane, boron trifluoride, aluminium chloride and alums.

Group - 14

Tendency for catenation; Structure, properties and uses of Allotropes and oxides of carbon, silicon tetrachloride, silicates, zeolites and silicones.

Group - 15

Properties and uses of nitrogen and phosphorus; Allotropic forms of phosphorus; Preparation, properties, structure and uses of ammonia, nitric acid, phosphine and phosphorus halides, (PCl_3 , PCl_5); Structures of oxides and oxoacids of nitrogen and phosphorus.

Group - 16

Preparation, properties, structures and uses of ozone; Allotropic forms of sulphur; Preparation, properties, structures and uses of sulphuric acid (including its industrial preparation); Structures of oxoacids of sulphur.

Group - 17

Preparation, properties and uses of hydrochloric acid; Trends in the acidic nature of hydrogen halides; Structures of Interhalogen compounds and oxides and oxoacids of halogens.

Group - 18

Occurrence and uses of noble gases; Structures of fluorides and oxides of xenon.

UNIT 16: d - and f - BLOCK ELEMENTS

Transition Elements

General introduction, electronic configuration, occurrence and characteristics, general trends in properties of the first row transition elements - physical properties, ionization enthalpy, oxidation states, atomic radii, colour, catalytic behaviour, magnetic properties, complex formation, interstitial compounds, alloy formation; Preparation, properties and uses of $\text{K}_2\text{Cr}_2\text{O}_7$ and KMnO_4 .

Inner Transition Elements

Lanthanoids - Electronic configuration, oxidation states and lanthanoid contraction.

Actinoids - Electronic configuration and oxidation states.

UNIT 17: CO-ORDINATION COMPOUNDS

Introduction to co-ordination compounds, Werner's theory; ligands, co-ordination number, denticity, chelation; IUPAC nomenclature of mononuclear co-ordination compounds, isomerism; Bonding-Valence bond approach and basic ideas of Crystal field theory, colour and magnetic properties; Importance of co-ordination compounds (in qualitative analysis, extraction of metals and in biological systems).

UNIT 18: ENVIRONMENTAL CHEMISTRY

Environmental pollution - Atmospheric, water and soil.

Atmospheric pollution - Tropospheric and Stratospheric

Tropospheric pollutants - Gaseous pollutants: Oxides of carbon, nitrogen and sulphur, hydrocarbons; their sources, harmful effects and prevention; Green house effect and Global warming; Acid rain;

Particulate pollutants: Smoke, dust, smog, fumes, mist; their sources, harmful effects and prevention.

Stratospheric pollution- Formation and breakdown of ozone, depletion of ozone layer - its mechanism and effects.

Water Pollution - Major pollutants such as, pathogens, organic wastes and chemical pollutants; their harmful effects and prevention.

Soil pollution - Major pollutants such as: Pesticides (insecticides, herbicides and fungicides), their harmful effects and prevention.

Strategies to control environmental pollution.

SECTION-C

ORGANIC CHEMISTRY

UNIT 19: PURIFICATION AND CHARACTERISATION OF ORGANIC COMPOUNDS

Purification - Crystallization, sublimation, distillation, differential extraction and chromatography - principles and their applications.

Qualitative analysis - Detection of nitrogen, sulphur, phosphorus and halogens.

Quantitative analysis (basic principles only) - Estimation of carbon, hydrogen, nitrogen, halogens, sulphur, phosphorus.

Calculations of empirical formulae and molecular formulae; Numerical problems in organic quantitative analysis.

UNIT 20: SOME BASIC PRINCIPLES OF ORGANIC CHEMISTRY

Tetravalency of carbon; Shapes of simple molecules - hybridization (s and p); Classification of organic compounds based on functional groups: and those containing halogens, oxygen, nitrogen and sulphur; Homologous series; Isomerism - structural and stereoisomerism.

Nomenclature (Trivial and IUPAC)

Covalent bond fission - Homolytic and heterolytic: free radicals, carbocations and carbanions; stability of carbocations and free radicals, electrophiles and nucleophiles.

Electronic displacement in a covalent bond

- Inductive effect, electromeric effect, resonance and hyperconjugation.

Common types of organic reactions- Substitution, addition, elimination and rearrangement.

UNIT 21: HYDROCARBONS

Classification, isomerism, IUPAC nomenclature, general methods of preparation, properties and reactions.

Alkanes - Conformations: Sawhorse and Newman projections (of ethane); Mechanism of halogenation of alkanes.

Alkenes - Geometrical isomerism; Mechanism of electrophilic addition: addition of hydrogen, halogens, water, hydrogen halides (Markownikoff's and peroxide effect); Ozonolysis and polymerization.

Alkynes - Acidic character; Addition of hydrogen, halogens, water and hydrogen halides; Polymerization.

Aromatic hydrocarbons - Nomenclature, benzene - structure and aromaticity; Mechanism of electrophilic substitution: halogenation, nitration, Friedel - Craft's alkylation and acylation, directive influence of functional group in mono-substituted benzene.

UNIT 22: ORGANIC COMPOUNDS CONTAINING HALOGENS

General methods of preparation, properties and reactions; Nature of C-X bond; Mechanisms of substitution reactions.

Uses; Environmental effects of chloroform, iodoform freons and DDT.

UNIT 23: ORGANIC COMPOUNDS CONTAINING OXYGEN

General methods of preparation, properties, reactions and uses.

ALCOHOLS, PHENOLS AND ETHERS

Alcohols: Identification of primary, secondary and tertiary alcohols; mechanism of dehydration.

Phenols: Acidic nature, electrophilic substitution reactions: halogenation, nitration and sulphonation, Reimer - Tiemann reaction.

Ethers: Structure.

Aldehyde and Ketones: Nature of carbonyl group; Nucleophilic addition to $>C=O$ group, relative reactivities of aldehydes and ketones; Important reactions such as - Nucleophilic addition reactions (addition of HCN, NH_3 and its derivatives), Grignard reagent; oxidation; reduction (Wolff Kishner and Clemmensen); acidity of α -hydrogen, aldol condensation, Cannizzaro reaction, Haloform reaction;

Chemical tests to distinguish between aldehydes and Ketones.

CARBOXYLIC ACIDS

Acidic strength and factors affecting it.

UNIT 24: ORGANIC COMPOUNDS CONTAINING NITROGEN

General methods of preparation, properties, reactions and uses.

Amines: Nomenclature, classification, structure, basic character and identification of primary, secondary and tertiary amines and their basic character.

Diazonium Salts: Importance in synthetic organic chemistry.

UNIT 25: POLYMERS

General introduction and classification of polymers, general methods of polymerization-addition and condensation, copolymerization;

Natural and synthetic rubber and vulcanization; some important polymers with emphasis on their monomers and uses - polythene, nylon, polyester and bakelite.

UNIT 26: BIOMOLECULES

General introduction and importance of biomolecules.

CARBOHYDRATES - Classification: aldoses and ketoses; monosaccharides (glucose and fructose) and constituent monosaccharides of oligosaccharides (sucrose, lactose and maltose).

PROTEINS - Elementary Idea of α -amino acids, peptide bond, polypeptides; Proteins: primary, secondary, tertiary and quaternary structure (qualitative idea only), denaturation of proteins, enzymes.

VITAMINS - Classification and functions.

NUCLEIC ACIDS - Chemical constitution of DNA and RNA.

Biological functions of nucleic acids.

UNIT 27: CHEMISTRY IN EVERYDAY LIFE

Chemicals in medicines - Analgesics, tranquilizers, antiseptics, disinfectants, antimicrobials, antifertility drugs, antibiotics, antacids, antihistamins - their meaning and common examples.

Chemicals in food - Preservatives, artificial sweetening agents - common examples.

Cleansing agents - Soaps and detergents, cleansing action.

UNIT 28: PRINCIPLES RELATED TO PRACTICAL CHEMISTRY

● Detection of extra elements (N,S, halogens) in organic compounds; Detection of the following

functional groups: hydroxyl (alcoholic and phenolic), carbonyl (aldehyde and ketone), carboxyl and amino groups in organic compounds.

● Chemistry involved in the preparation of the following:

Inorganic compounds: Mohr's salt, potash alum.

Organic compounds: Acetanilide, p-nitroacetanilide, aniline yellow, iodoform.

● Chemistry involved in the titrimetric exercises - Acids bases and the use of indicators, oxalic-acid vs KMnO_4 , Mohr's salt vs KMnO_4 .

● Chemical principles involved in the qualitative salt analysis:

Cations - Pb^{2+} , Cu^{2+} , Al^{3+} , Fe^{3+} , Zn^{2+} , Ni^{2+} , Ca^{2+} , Ba^{2+} , Mg^{2+} , NH_4^+ .

Anions- CO_3^{2-} , S^{2-} , SO_4^{2-} , NO_3^- , NO_2^- , Cl^- , Br^- , I^- . (Insoluble salts excluded).

● Chemical principles involved in the following experiments:

1. Enthalpy of solution of CuSO_4
2. Enthalpy of neutralization of strong acid and strong base.
3. Preparation of lyophilic and lyophobic sols.
4. Kinetic study of reaction of iodide ion with hydrogen peroxide at room temperature.

SYLLABUS FOR APTITUDE TEST B. ARCH/B. PLANNING

Part - I Awareness of persons, places, Buildings, Materials.) Objects, Texture related to Architecture and build-environment. Visualising three dimensional objects from two dimensional drawings. Visualising different sides of three dimensional objects. Analytical Reasoning Mental Ability (Visual, Numerical and Verbal).

Part - II Three dimensional - perception: Understanding and appreciation of scale and proportion of objects, building forms and elements, colour texture, harmony and contrast. Design and drawing of geometrical or abstract shapes and patterns in pencil. Transformation of forms both 2 D and 3 D union, subtraction, rotation, development of surfaces and volumes, Generation of Plan, elevations and 3 D views of objects. Creating two dimensional and three dimensional compositions using given shapes and forms.

Sketching of scenes and activities from memory of urbanscape (public space, market, festivals, street scenes, monuments, recreational spaces etc.), landscape (river fronts, jungles, gardens, trees, plants etc.) and rural life.

Note: Candidates are advised to bring pencils, own geometry box set, erasers and colour pencils and crayons for the Aptitude Test.

BIOLOGY

CONTENTS OF CLASS XI SYLLABUS

UNIT I: Diversity in Living World

- What is living? ; Biodiversity; Need for classification; Three domains of life; Taxonomy & Systematics; Concept of species and taxonomical hierarchy; Binomial nomenclature; Tools for study of Taxonomy – Museums, Zoos, Herbaria, Botanical gardens.
- Five kingdom classification; salient features and classification of Monera; Protista and Fungi into major groups; Lichens; Viruses and Viroids.
- Salient features and classification of plants into major groups-Algae, Bryophytes, Pteridophytes, Gymnosperms and Angiosperms (three to five salient and distinguishing features and at least two examples of each category); Angiosperms- classification up to class, characteristic features and examples).
- Salient features and classification of animals-nonchordate up to phyla level and chordate up to classes level (three to five salient features and at least two examples).

UNIT II: Structural Organisation in Animals and Plants

- Morphology and modifications; Tissues; Anatomy and functions of different parts of flowering plants: Root, stem, leaf, inflorescence- cymose and racemose, flower, fruit and seed (To be dealt along with the relevant practical of the Practical Syllabus).
- Animal tissues; Morphology, anatomy and functions of different systems (digestive, circulatory, respiratory, nervous and reproductive) of an insect (cockroach). (Brief account only)

UNIT III: Cell Structure and Function

- Cell theory and cell as the basic unit of life; Structure of prokaryotic and eukaryotic cell; Plant cell and animal cell; Cell envelope, cell membrane, cell wall; Cell organelles-structure and function; Endomembrane system-endoplasmic reticulum, Golgi bodies, lysosomes, vacuoles; mitochondria, ribosomes, plastids, micro bodies; Cytoskeleton, cilia, flagella, centrioles (ultrastructure and function); Nucleus-nuclear membrane, chromatin, nucleolus.
- Chemical constituents of living cells: Biomolecules-structure and function of proteins, carbohydrates, lipids, nucleic acids; Enzymes-types, properties, enzyme action.
- B Cell division: Cell cycle, mitosis, meiosis and their significance.

UNIT IV: Plant Physiology

- Transport in plants: Movement of water, gases and nutrients; Cell to cell transport-Diffusion, facilitated diffusion, active transport; Plant – water relations – Imbibition, water potential, osmosis, plasmolysis; Long distance transport of water – Absorption, apoplast, symplast, transpiration pull, root pressure and guttation; Transpiration-Opening and closing of stomata; Uptake and translocation of mineral nutrients-Transport of food, phloem transport, Mass flow hypothesis; Diffusion of gases (brief mention).
- Mineral nutrition: Essential minerals, macro and micronutrients and their role; Deficiency symptoms; Mineral toxicity; Elementary idea of Hydroponics as a method to study mineral nutrition; Nitrogen metabolism-Nitrogen cycle, biological nitrogen fixation.
- Photosynthesis: Photosynthesis as a means of Autotrophic nutrition; Site of photosynthesis take place; pigments involved in Photosynthesis (Elementary idea); Photochemical and biosynthetic phases of photosynthesis; Cyclic and noncyclic and photophosphorylation; Chemiosmotic hypothesis; Photorespiration C₃ and C₄ pathways; Factors affecting photosynthesis.
- Respiration: Exchange gases; Cellular respiration-glycolysis, fermentation (anaerobic), TCA cycle and electron transport system (aerobic); Energy relations-Number of ATP molecules generated; Amphibolic pathways; Respiratory quotient.
- Plant growth and development: Seed germination; Phases of Plant growth and plant growth rate; Conditions of growth; Differentiation, dedifferentiation and redifferentiation; Sequence of developmental process in a plant cell; Growth regulators-auxin,gibberellin, cytokinin, ethylene, ABA; Seed dormancy; Vernalisation;

UNIT IV: Human Physiology

- Digestion and absorption; Alimentary canal and digestive glands; Role of digestive enzymes and gastrointestinal hormones; Peristalsis, digestion, absorption and assimilation of proteins, carbohydrates and fats; Caloric value of proteins, carbohydrates and fats; Egestion; Nutritional and digestive disorders – PEM, indigestion, constipation, vomiting, jaundice, diarrhea.
- Breathing and Respiration: Respiratory organs in animals (recall only); Respiratory system in humans; Mechanism of breathing and its regulation in humans-Exchange of gases, transport of gases and regulation of respiration Respiratory volumes; Disorders related to respiration-Asthma, Emphysema, Occupational respiratory disorders.
- Body fluids and circulation: Composition of blood, blood groups, coagulation of blood; Composition of lymph and its function; Human circulatory system-Structure of human heart and blood vessels; Cardiac cycle, cardiac output, ECG, Double circulation; Regulation of cardiac activity; Disorders of circulatory system-Hypertension, Coronary artery disease, Angina pectoris, Heart failure.
- Excretory products and their elimination: Modes of excretion- Ammonotelism, ureotelism, uricotelism; Human excretory system-structure and fuction; Urine formation, Osmoregulation;

Regulation of kidney function-Renin-angiotensin, Atrial Natriuretic Factor, ADH and Diabetes insipidus; Role of other organs in excretion; Disorders; Uraemia, Renal failure, Renal calculi, Nephritis; Dialysis and artificial kidney.

- Locomotion and Movement: Types of movement- ciliary, flagellar, muscular; Skeletal muscle-contractile proteins and muscle contraction; Skeletal system and its functions (To be dealt with the relevant practical of Practical syllabus); Joints; Disorders of muscular and skeletal system- Myasthenia gravis, Tetany, Muscular dystrophy, Arthritis, Osteoporosis, Gout.

- Neural control and coordination: Neuron and nerves; Nervous system in humans- central nervous system, peripheral nervous system and visceral nervous system; Generation and conduction of nerve impulse; Reflex action; Sense organs; Elementary structure and function of eye and ear.

- Chemical coordination and regulation: Endocrine glands and hormones; Human endocrine system-Hypothalamus, Pituitary, Pineal, Thyroid, Parathyroid, Adrenal, Pancreas, Gonads; Mechanism of hormone action (Elementary Idea); Role of hormones as messengers and regulators, Hypo-and hyperactivity and related disorders (Common disorders e.g. Dwarfism, Acromegaly, Cretinism, goiter, exophthalmic goiter, diabetes, Addison's disease).

(Imp: Diseases and disorders mentioned above to be dealt in brief.)

CONTENTS OF CLASS XII SYLLABUS

UNIT I: Reproduction

- Reproduction in organisms: Reproduction, a characteristic feature of all organisms for continuation of species; Modes of reproduction – Asexual and sexual; Asexual reproduction; Modes-Binary fission, sporulation, budding, gemmule, fragmentation; vegetative propagation in plants.

- Sexual reproduction in flowering plants: Flower structure; Development of male and female gametophytes; Pollination-types, agencies and examples; Outbreeding devices; Pollen-Pistil interaction; Double fertilization; Post fertilization events-Development of endosperm and embryo, Development of seed and formation of fruit; Special modes-apomixis, parthenocarpy, polyembryony; Significance of seed and fruit formation.

- Human Reproduction: Male and female reproductive systems; Microscopic anatomy of testis and ovary; Gametogenesis-spermatogenesis & oogenesis; Menstrual cycle; Fertilisation, embryo development upto blastocyst formation, implantation; Pregnancy and placenta formation (Elementary idea); Parturition (Elementary idea); Lactation (Elementary idea).

- Reproductive health: Need for reproductive health and prevention of sexually transmitted diseases (STD); Birth control-Need and Methods, Contraception and Medical Termination of Pregnancy (MTP); Amniocentesis; Infertility and assisted reproductive technologies – IVF, ZIFT, GIFT (Elementary idea for general awareness).

UNIT II: Genetics and Evolution

- Heredity and variation: Mendelian Inheritance; Deviations from Mendelism-Incomplete dominance, Co-dominance, Multiple alleles and Inheritance of blood groups, Pleiotropy; Elementary idea of polygenic inheritance; Chromosome theory of inheritance; Chromosomes and genes; Sex determination-In humans, birds, honey bee; Linkage and crossing over; Sex linked inheritance-Haemophilia, Colour blindness; Mendelian disorders in humans-Thalassemia; Chromosomal disorders in humans; Down's syndrome, Turner's and Klinefelter's syndromes.
- Molecular basis of Inheritance: Search for genetic material and DNA as genetic material; Structure of DNA and RNA; DNA packaging; DNA replication; Central dogma; Transcription, genetic code, translation; Gene expression and regulation-Lac Operon; Genome and human genome project; DNA finger printing.
- Evolution: Origin of life; Biological evolution and evidences for biological evolution from Paleontology, comparative anatomy, embryology and molecular evidence); Darwin's contribution, Modern Synthetic theory of Evolution; Mechanism of evolution-Variation (Mutation and Recombination) and Natural Selection with examples, types of natural selection; Gene flow and genetic drift; Hardy-Weinberg's principle; Adaptive Radiation; Human evolution.

UNIT III: Biology and Human Welfare

- Health and Disease; Pathogens; parasites causing human diseases (Malaria, Filariasis, Ascariasis, Typhoid, Pneumonia, common cold, amoebiasis, ring worm); Basic concepts of immunology-vaccines; Cancer, HIV and AIDS; Adolescence, drug and alcohol abuse.
- Improvement in food production; Plant breeding, tissue culture, single cell protein, Biofortification; Apiculture and Animal husbandry.
- Microbes in human welfare: In household food processing, industrial production, sewage treatment, energy generation and as biocontrol agents and biofertilizers.

UNIT IV: Biotechnology and Its Applications

- Principles and process of Biotechnology: Genetic engineering (Recombinant DNA technology).
- Application of Biotechnology in health and agriculture: Human insulin and vaccine production, gene therapy; Genetically modified organisms-Bt crops; Transgenic Animals; Biosafety issues-Biopiracy and patents.

UNIT V: Ecology and environment

- Organisms and environment: Habitat and niche; Population and ecological adaptations; Population interactions-mutualism, competition, predation, parasitism; Population attributes-growth, birth rate and death rate, age distribution.

- Ecosystem: Patterns, components; productivity and decomposition; Energy flow; Pyramids of number, biomass, energy; Nutrient cycling (carbon and phosphorous); Ecological succession; Ecological Services-Carbon fixation, pollination, oxygen release.
- Biodiversity and its conservation: Concept of Biodiversity; Patterns of Biodiversity; Importance of Biodiversity; Loss of Biodiversity; Biodiversity conservation; Hotspots, endangered organisms, extinction, Red Data Book, biosphere reserves, National parks and sanctuaries.
- Environmental issues: Air pollution and its control; Water pollution and its control; Agrochemicals and their effects; Solid waste management; Radioactive waste management; Greenhouse effect and global warming; Ozone depletion; Deforestation; Any three case studies as success stories addressing environmental issues.

Syllabus for M.Tech

SECTION- A

(Common to all branches)

General Aptitude

- (a) Verbal Ability: English grammar, sentence completion, verbal analogies, word groups, instructions, critical reasoning and verbal deduction.
- (b) Numerical Ability: Numerical computation, numerical estimation, numerical reasoning and data interpretation.

SECTION- B

Branch wise Syllabus for M.Tech.

Civil Engineering (CE)

(a) *Engineering Mathematics*

Linear Algebra: Matrix algebra, Systems of linear equations, Eigen values and eigenvectors.

Calculus: Functions of single variable, Limit, continuity and differentiability, Mean value theorems, Evaluation of definite and improper integrals, Partial derivatives, Total derivative, Maxima and minima, Gradient, Divergence and Curl, Vector identities, Directional derivatives, Line, Surface and Volume integrals, Stokes, Gauss and Green's theorems.

Differential equations: First order equations (linear and nonlinear), Higher order linear differential equations with constant coefficients, Cauchy's and Euler's equations, Initial and boundary value problems, Laplace transforms, Solutions of one dimensional heat and wave equations and Laplace equation.

Complex variables: Analytic functions, Cauchy's integral theorem, Taylor and Laurent series. Probability and Statistics: Definitions of probability and sampling theorems, Conditional probability, Mean, median, mode and standard deviation, Random variables, Poisson, Normal and Binomial distributions.

Numerical Methods: Numerical solutions of linear and non-linear algebraic equations
Integration by trapezoidal and Simpson's rule, single and multi-step methods for differential equations.

(b) *Structural Engineering*

Mechanics: Bending moment and shear force in statically determinate beams. Simple stress and strain relationship: Stress and strain in two dimensions, principal stresses, stress transformation, Mohr's circle. Simple bending theory, flexural and shear stresses, unsymmetrical bending, shear centre. Thin walled pressure vessels, uniform torsion, buckling of column, combined and direct bending stresses.

Structural Analysis: Analysis of statically determinate trusses, arches, beams, cables and frames, displacements in statically determinate structures and analysis of statically indeterminate structures by force/ energy methods, analysis by displacement methods (slope deflection and moment distribution methods), influence lines for determinate and indeterminate structures. Basic concepts of matrix methods of structural analysis.

Concrete Structures: Concrete Technology- properties of concrete, basics of mix design. Concrete design-basic working stress and limit state design concepts, analysis of ultimate load capacity and design of members subjected to flexure, shear, compression and torsion by limit state methods. Basic elements of prestressed concrete, analysis of beam sections at transfer and service loads.

Steel Structures: Analysis and design of tension and compression members, beams and beam- columns, column bases. Connections- simple and eccentric, beam-column connections, plate girders and trusses. Plastic analysis of beams and frames.

(c) *Geotechnical Engineering*

Soil Mechanics: Origin of soils, soil classification, three-phase system, fundamental definitions, relationship and interrelationships, permeability & seepage, effective stress principle, consolidation, compaction, shear strength.

Foundation Engineering: Sub-surface investigations- scope, drilling bore holes, sampling, penetration tests, plate load test. Earth pressure theories, effect of water table, layered soils. Stability of slopes-infinite slopes, finite slopes. Foundation types-foundation design requirements. Shallow foundations-bearing capacity, effect of shape, water table and other factors, stress distribution, settlement analysis in sands & clays. Deep foundations-pile types, dynamic & static formulae, load capacity of piles in sands & clays, negative skin friction.

(d) *Water Resources Engineering*

Fluid Mechanics and Hydraulics: Properties of fluids, principle of conservation of mass, momentum, energy and corresponding equations, potential flow, applications of momentum and Bernoulli's equation, laminar and turbulent flow, flow in pipes, pipe networks. Concept of boundary layer and its growth. Uniform flow, critical flow and gradually varied flow in channels, specific energy concept, hydraulic jump. Forces on immersed bodies, flow measurements in channels, tanks and pipes. Dimensional analysis and hydraulic modeling. Kinematics of flow, velocity triangles and specific speed of pumps and turbines.

Hydrology: Hydrologic cycle, rainfall, evaporation, infiltration, stage discharge relationships, unit hydrographs, flood estimation, reservoir capacity, reservoir and channel routing. Well hydraulics.

Irrigation: Duty, delta, estimation of evapo-transpiration. Crop water requirements. Design of: lined and unlined canals, waterways, head works, gravity dams and spillways. Design of weirs on permeable foundation. Types of irrigation system, irrigation methods. Water logging and drainage, sodic soils.

(e) *Environmental Engineering*

Water requirements: Quality standards, basic unit processes and operations for water treatment. Drinking water standards, water requirements, basic unit operations and unit processes for surface water treatment, distribution of water. Sewage and sewerage treatment, quantity and characteristics of wastewater. Primary, secondary and tertiary treatment of wastewater, sludge disposal, effluent discharge standards. Domestic wastewater treatment, quantity of characteristics of domestic wastewater, primary and secondary treatment Unit operations and unit processes of domestic wastewater, sludge disposal.

Air Pollution: Types of pollutants, their sources and impacts, air pollution meteorology, air pollution control, air quality standards and limits.

Municipal Solid Wastes: Characteristics, generation, collection and transportation of solid wastes, engineered systems for solid waste management (reuse/ recycle, energy recovery, treatment and disposal).

Noise Pollution: Impacts of noise, permissible limits of noise pollution, measurement of noise and control of noise pollution.

(f) *Transportation Engineering*

Highway Planning: Geometric design of highways, testing and specifications of paving materials, design of flexible and rigid pavements.

Traffic Engineering: Traffic characteristics, theory of traffic flow, intersection design, traffic signs and signal design, highway capacity.

(g) ***Surveying***

Importance of surveying, principles and classifications, mapping concepts, coordinate system, map projections, measurements of distance and directions, leveling, theodolite traversing, plane table surveying, errors and adjustments, curves.

Computer Science and Information Technology (CS)

(a) ***Engineering Mathematics***

Mathematical Logic: Propositional Logic; First Order Logic.

Probability: Conditional Probability; Mean, Median, Mode and Standard Deviation; Random Variables; Distributions; uniform, normal, exponential, Poisson, Binomial.

Set Theory & Algebra: Sets; Relations; Functions; Groups; Partial Orders; Lattice; Boolean Algebra.

Combinatorics: Permutations; Combinations; Counting; Summation; generating functions; recurrence relations; asymptotics.

Graph Theory: Connectivity; spanning trees; Cut vertices & edges; covering; matching; independent sets; Colouring; Planarity; Isomorphism.

Linear Algebra: Algebra of matrices, determinants, systems of linear equations, Eigen values and Eigen vectors.

Numerical Methods: LU decomposition for systems of linear equations; numerical solutions of nonlinear algebraic equations by Secant, Bisection and Newton-Raphson Methods; Numerical integration by trapezoidal and Simpson's rules.

Calculus: Limit, Continuity & differentiability, Mean value Theorems, Theorems of integral calculus, evaluation of definite & improper integrals, Partial derivatives, Total derivatives, maxima & minima.

(b) ***Computer Science and Information Technology***

Digital Logic: Logic functions, Minimization, Design and synthesis of combinational and sequential circuits; Number representation and computer arithmetic (fixed and floating point).

Computer Organization and Architecture: Machine instructions and addressing modes, ALU and datapath, CPU control design, Memory interface, I/O interface (Interrupt and DMA mode), Instruction pipelining, Cache and main memory, Secondary storage.

Programming and Data Structures: Programming in C; Functions, Recursion, Parameter passing, Scope, Binding; Abstract data types, Arrays, Stacks, Queues, Linked Lists, Trees, Binary search trees, Binary heaps.

Algorithms: Analysis, Asymptotic notation, Notions of space and time complexity, Worst and average case analysis; Design: Greedy approach, Dynamic programming, Divide-and-conquer; Tree and graph traversals, Connected components, Spanning trees, Shortest paths; Hashing, Sorting, Searching. Asymptotic analysis (best, worst, average cases) of time and space, upper and lower bounds, Basic concepts of complexity classes – P, NP, NP-hard, NP-complete.

Theory of Computation: Regular languages and finite automata, Context free languages and Push-down automata, Recursively enumerable sets and Turing machines, Undecidability.

Compiler Design: Lexical analysis, Parsing, Syntax directed translation, Runtime environments, Intermediate and target code generation, Basics of code optimization.

Operating System: Processes, Threads, Inter-process communication, Concurrency, Synchronization, Deadlock, CPU scheduling, Memory management and virtual memory, File systems, I/O systems, Protection and security.

Databases: ER-model, Relational model (relational algebra, tuple calculus), Database design (integrity constraints, normal forms), Query languages (SQL), File structures (sequential files, indexing, B and B+trees), Transactions and concurrency control.

Information Systems and Software Engineering: information gathering, requirement and feasibility analysis, data flow diagrams, process specifications, input/output design, process life cycle, planning and managing the project, design, coding, testing, implementation, maintenance.

Computer Networks: ISO/OSI stack, LAN technologies (Ethernet, Token ring), Flow and error control techniques, Routing algorithms, Congestion control, TCP/UDP and sockets, IP(v4), Application layer protocols (icmp, dns, smtp, pop, ftp, http); Basic concepts of hubs,

switches, gateways, and routers. Network security – basic concepts of public key and private key cryptography, digital signature, firewalls.

Web technologies: HTML, XML, basic concepts of client-server computing.

Electronics and Communication Engineering (EC)

(a) *Engineering Mathematics*

Linear Algebra: Matrix Algebra, Systems of linear equations, Eigen values and eigen vectors.

Calculus: Mean value theorems, Theorems of integral calculus, Evaluation of definite and improper integrals, Partial Derivatives, Maxima and minima, Multiple integrals, Fourier series. Vector identities, Directional derivatives, Line, Surface and Volume integrals, Stokes, Gauss and Green's theorems.

Differential equations: First order equation (linear and nonlinear), Higher order linear differential equations with constant coefficients, Method of variation of parameters, Cauchy's and Euler's equations, Initial and boundary value problems, Partial Differential Equations and variable separable method.

Complex variables: Analytic functions, Cauchy's integral theorem and integral formula, Taylor's and Laurent' series, Residue theorem, solution integrals.

Probability and Statistics: Sampling theorems, Conditional probability, Mean, median, mode and standard deviation, Random variables, Discrete and continuous distributions, Poisson, Normal and Binomial distribution, Correlation and regression analysis.

Numerical Methods: Solutions of non-linear algebraic equations, single and multi-step methods for differential equations.

Transform Theory: Fourier transform, Laplace transform, Z-transform.

(b) *Electronics and Communication Engineering*

Networks: Network graphs: matrices associated with graphs; incidence, fundamental cut set and fundamental circuit matrices. Solution methods: nodal and mesh analysis. Network theorems: superposition, Thevenin and Norton's maximum power transfer, Wye-Delta transformation. Steady state sinusoidal analysis using phasors. Linear constant coefficient differential equations; time domain analysis of simple RLC circuits, Solution of network

equations using Laplace transform: frequency domain analysis of RLC circuits. 2-port network parameters: driving point and transfer functions. State equations for networks.

Electronic Devices: Energy bands in silicon, intrinsic and extrinsic silicon. Carrier transport in silicon: diffusion current, drift current, mobility, and resistivity. Generation and recombination of carriers. p-n junction diode, Zener diode, tunnel diode, BJT, JFET, MOS capacitor, MOSFET, LED, p-I-n and avalanche photo diode, Basics of LASERS. Device technology: integrated circuits fabrication process, oxidation, diffusion, ion implantation, photolithography, n-tub, p-tub and twin-tub CMOS process.

Analog Circuits: Small Signal Equivalent circuits of diodes, BJTs, MOSFETs and analog CMOS. Simple diode circuits, clipping, clamping, rectifier. Biasing and bias stability of transistor and FET amplifiers. Amplifiers: single- and multi-stage, differential and operational, feedback, and power. Frequency response of amplifiers. Simple op-amp circuits. Filters. Sinusoidal oscillators; criterion for oscillation; single-transistor and op-amp configurations. Function generators and wave-shaping circuits, 555 Timers. Power supplies.

Digital circuits: Boolean algebra, minimization of Boolean functions; logic gates; digital IC families (DTL, TTL, ECL, MOS, CMOS). Combinatorial circuits: arithmetic circuits, code converters, multiplexers, decoders, PROMs and PLAs. Sequential circuits: latches and flip-flops, counters and shift registers. Sample and hold circuits, ADCs, DACs. Semiconductor memories. Microprocessor(8085): architecture, programming, memory and I/O interfacing.

Signals and Systems: Definitions and properties of Laplace transform, continuous-time and discrete-time Fourier series, continuous-time and discrete-time Fourier Transform, DFT and FFT, z-transform.

Sampling theorem. Linear Time-Invariant (LTI) Systems: definitions and properties; causality, stability, impulse response, convolution, poles and zeros, parallel and cascade structure, frequency response, group delay, phase delay. Signal transmission through LTI systems.

Control Systems: Basic control system components; block diagrammatic description, reduction of block diagrams. Open loop and closed loop (feedback) systems and stability analysis of these systems. Signal flow graphs and their use in determining transfer functions of systems; transient and steady state analysis of LTI control systems and frequency response. Tools and techniques for LTI control system analysis: root loci, Routh-Hurwitz criterion, Bode and Nyquist plots. Control system compensators: elements of lead and lag compensation, elements of Proportional-Integral-Derivative (PID) control. State variable representation and solution of state equation of LTI control systems.

Communications: Random signals and noise: probability, random variables, probability density function, autocorrelation, power spectral density. Analog communication systems: amplitude and angle modulation and demodulation systems, spectral analysis of these operations, superheterodyne receivers; elements of hardware, realizations of analog communication systems; signal-to-noise ratio (SNR) calculations for amplitude modulation (AM) and frequency modulation (FM) for low noise conditions. Fundamentals of information theory and channel capacity theorem. Digital communication systems: pulse code modulation (PCM), differential pulse code modulation (DPCM), digital modulation schemes: amplitude, phase and frequency shift keying schemes (ASK, PSK, FSK), matched filter receivers, bandwidth consideration and probability of error calculations for these schemes. Basics of TDMA, FDMA and CDMA and GSM.

Electromagnetics: Elements of vector calculus: divergence and curl; Gauss' and Stokes' theorems, Maxwell's equations: differential and integral forms. Wave equation, Poynting vector. Plane waves: propagation through various media; reflection and refraction; phase and group velocity; skin depth. Transmission lines: characteristic impedance; impedance transformation; Smith chart; impedance matching; S parameters, pulse excitation. Waveguides: modes in rectangular waveguides; boundary conditions; cut-off frequencies; dispersion relations. Basics of propagation in dielectric waveguide and optical fibers. Basics of Antennas: Dipole antennas; radiation pattern; antenna gain.

Electrical Engineering (EE)

(a) *Engineering Mathematics*

Linear Algebra: Matrix Algebra, Systems of linear equations, Eigen values and eigen vectors.

Calculus: Mean value theorems, Theorems of integral calculus, Evaluation of definite and improper integrals, Partial Derivatives, Maxima and minima, Multiple integrals, Fourier series. Vector identities, Directional derivatives, Line, Surface and Volume integrals, Stokes, Gauss and Green's theorems.

Differential equations: First order equation (linear and nonlinear), Higher order linear differential equations with constant coefficients, Method of variation of parameters, Cauchy's and Euler's equations, Initial and boundary value problems, Partial Differential Equations and variable separable method.

Complex variables: Analytic functions, Cauchy's integral theorem and integral formula, Taylor's and Laurent' series, Residue theorem, solution integrals.

Probability and Statistics: Sampling theorems, Conditional probability, Mean, median, mode and standard deviation, Random variables, Discrete and continuous distributions, Poisson, Normal and Binomial distribution, Correlation and regression analysis.

Numerical Methods: Solutions of non-linear algebraic equations, single and multi-step methods for differential equations.

Transform Theory: Fourier transform, Laplace transform, Z-transform.

(b) ***Electrical Engineering***

Electric Circuits and Fields: Network graph, KCL, KVL, node and mesh analysis, transient response of dc and ac networks; sinusoidal steady-state analysis, resonance, basic filter concepts; ideal current and voltage sources, Thevenin's, Norton's and Superposition and Maximum Power Transfer theorems, twoport networks, three phase circuits; Gauss Theorem, electric field and potential due to point, line, plane and spherical charge distributions; Ampere's and Biot-Savart's laws; inductance; dielectrics; capacitance.

Signals and Systems: Representation of continuous and discrete-time signals; shifting and scaling operations; linear, time-invariant and causal systems; Fourier series representation of continuous periodic signals; sampling theorem; Fourier, Laplace and Z transforms.

Electrical Machines: Single phase transformer – equivalent circuit, phasor diagram, tests, regulation and efficiency; three phase transformers – connections, parallel operation; auto-transformer; energy conversion principles; DC machines – types, windings, generator characteristics, armature reaction and commutation, starting and speed control of motors; three phase induction motors – principles, types, performance characteristics, starting and speed control; single phase induction motors; synchronous machines – performance, regulation and parallel operation of generators, motor starting, characteristics and applications; servo and stepper motors.

Power Systems: Basic power generation concepts; transmission line models and performance; cable performance, insulation; corona and radio interference; distribution systems; per-unit quantities; bus impedance and admittance matrices; load flow; voltage control; power factor correction; economic operation; symmetrical components; fault analysis; principles of over-current, differential and distance protection; solid state relays and digital protection; circuit breakers; system stability concepts, swing curves and equal area criterion; HVDC transmission and FACTS concepts.

Control Systems: Principles of feedback; transfer function; block diagrams; steady-state errors; Routh and Niquist techniques; Bode plots; root loci; lag, lead and lead-lag compensation; state space model; state transition matrix, controllability and observability.

Electrical and Electronic Measurements: Bridges and potentiometers; PMMC, moving iron, dynamometer and induction type instruments; measurement of voltage, current, power, energy and power factor; instrument transformers; digital voltmeters and multimeters; phase, time and frequency measurement; Q-meters; oscilloscopes; potentiometric recorders; error analysis.

Analog and Digital Electronics: Characteristics of diodes, BJT, FET; amplifiers – biasing, equivalent circuit and frequency response; oscillators and feedback amplifiers; operational amplifiers—characteristics and applications; simple active filters; VCOs and timers; combinational and sequential logic circuits; multiplexer; Schmitt trigger; multi-vibrators; sample and hold circuits; A/D and D/A converters; 8-bit microprocessor basics, architecture, programming and interfacing.

Power Electronics and Drives: Semiconductor power diodes, transistors, thyristors, triacs, GTOs, MOSFETs and IGBTs – static characteristics and principles of operation; triggering circuits; phase control rectifiers; bridge converters – fully controlled and half controlled; principles of choppers and inverters; basic concepts of adjustable speed dc and ac drives.

Mechanical Engineering (ME)

(a) *Engineering Mathematics*

Linear Algebra: Matrix algebra, Systems of linear equations, Eigen values and eigen vectors.

Calculus: Functions of single variable, Limit, continuity and differentiability, Mean value theorems, Evaluation of definite and improper integrals, Partial derivatives, Total derivative, Maxima and minima, Gradient, Divergence and Curl, Vector identities, Directional derivatives, Line, Surface and Volume integrals, Stokes, Gauss and Green's theorems.

Differential equations: First order equations (linear and nonlinear), Higher order linear differential equations with constant coefficients, Cauchy's and Euler's equations, Initial and boundary value problems, Laplace transforms, Solutions of one dimensional heat and wave equations and Laplace equation.

Complex variables: Analytic functions, Cauchy's integral theorem, Taylor and Laurent series.

Probability and Statistics: Definitions of probability and sampling theorems, Conditional probability, Mean, median, mode and standard deviation, Random variables, Poisson, Normal and Binomial distributions.

Numerical Methods: Numerical solutions of linear and non-linear algebraic equations
Integration by trapezoidal and Simpson's rule, single and multi-step methods for differential equations.

(b) *Applied Mechanics And Design*

Engineering Mechanics: Free body diagrams and equilibrium; trusses and frames; virtual work; kinematics and dynamics of particles and of rigid bodies in plane motion, including impulse and momentum (linear and angular) and energy formulations; impact.

Strength of Materials: Stress and strain, stress-strain relationship and elastic constants, Mohr's circle for plane stress and plane strain, thin cylinders; shear force and bending moment diagrams; bending and shear stresses; deflection of beams; torsion of circular shafts; Euler's theory of columns; strain energy methods; thermal stresses.

Theory of Machines: Displacement, velocity and acceleration analysis of plane mechanisms; dynamic analysis of slider-crank mechanism; gear trains; flywheels.

Vibrations: Free and forced vibration of single degree of freedom systems; effect of damping; vibration isolation; resonance, critical speeds of shafts.

Design: Design for static and dynamic loading; failure theories; fatigue strength and the S-N diagram; *principles* of the design of machine elements such as bolted, riveted and welded joints, shafts, spur gears, rolling and sliding contact bearings, brakes and clutches.

(c) *Fluid Mechanics And Thermal Sciences*

Fluid Mechanics: Fluid properties; fluid statics, manometry, buoyancy; control-volume analysis of mass, momentum and energy; fluid acceleration; differential equations of continuity and momentum; Bernoulli's equation; viscous flow of incompressible fluids; boundary layer; elementary turbulent flow; flow through pipes, head losses in pipes, bends etc.

Heat-Transfer: Modes of heat transfer; one dimensional heat conduction, resistance concept, electrical analogy, unsteady heat conduction, fins; dimensionless parameters in free and forced convective heat transfer, various correlations for heat transfer in flow over flat plates and through pipes; thermal boundary layer; effect of turbulence; radiative heat transfer, black and grey surfaces, shape factors, network analysis; heat exchanger performance, LMTD and NTU methods.

Thermodynamics: Zeroth, First and Second laws of thermodynamics; thermodynamic system and processes; Carnot cycle. irreversibility and availability; behaviour of ideal and

real gases, properties of pure substances, calculation of work and heat in ideal processes; analysis of thermodynamic cycles related to energy conversion.

Applications: *Power Engineering:* Steam Tables, Rankine, Brayton cycles with regeneration and reheat. *I.C. Engines:* air-standard Otto, Diesel cycles. *Refrigeration and air-conditioning:* Vapour refrigeration cycle, heat pumps, gas refrigeration, Reverse Brayton cycle; moist air: psychrometric chart, basic psychrometric processes. *Turbomachinery:* Pelton-wheel, Francis and Kaplan turbines — impulse and reaction principles, velocity diagrams.

(d) ***Manufacturing And Industrial Engineering***

Engineering Materials: Structure and properties of engineering materials, heat treatment, stress-strain diagrams for engineering materials.

Metal Casting: Design of patterns, moulds and cores; solidification and cooling; riser and gating design, design considerations.

Forming: Plastic deformation and yield criteria; fundamentals of hot and cold working processes; load estimation for bulk (forging, rolling, extrusion, drawing) and sheet (shearing, deep drawing, bending) metal forming processes; principles of powder metallurgy.

Joining: Physics of welding, brazing and soldering; adhesive bonding; design considerations in welding.

Machining and Machine Tool Operations: Mechanics of machining, single and multi-point cutting tools, tool geometry and materials, tool life and wear; economics of machining; principles of nontraditional machining processes; principles of work holding, principles of design of jigs and fixtures.

Metrology and Inspection: Limits, fits and tolerances; linear and angular measurements; comparators; gauge design; interferometry; form and finish measurement; alignment and testing methods; tolerance analysis in manufacturing and assembly.

Computer Integrated Manufacturing: Basic concepts of CAD/CAM and their integration tools.

Production Planning and Control: Forecasting models, aggregate production planning, scheduling, materials requirement planning.

Inventory Control: Deterministic and probabilistic models; safety stock inventory control systems.

Operations Research: Linear programming, simplex and duplex method, transportation, assignment, network flow models, simple queuing models, PERT and CPM

SECTION- C
(Common to all branches)
Engineering Mathematics

Linear Algebra: Algebra of matrices, inverse, rank, system of linear equations, symmetric, skewsymmetric and orthogonal matrices. Hermitian, skew-Hermitian and unitary matrices. eigenvalues and eigenvectors, diagonalisation of matrices, Cayley-Hamilton Theorem.

Calculus: Functions of single variable, limit, continuity and differentiability, Mean value theorems, Indeterminate forms and L'Hospital rule, Maxima and minima, Taylor's series, Fundamental and mean value-theorems of integral calculus. Evaluation of definite and improper integrals, Beta and Gamma functions, Functions of two variables, limit, continuity, partial derivatives, Euler's theorem for homogeneous functions, total derivatives, maxima and minima, Lagrange method of multipliers, double and triple integrals and their applications, sequence and series, tests for convergence, power series, Fourier Series, Half range sine and cosine series.

Complex variable: Analytic functions, Cauchy-Riemann equations, Application in solving potential problems, Line integral, Cauchy's integral theorem and integral formula (without proof), Taylor's and Laurent's series, Residue theorem (without proof) and its applications.

Vector Calculus: Gradient, divergence and curl, vector identities, directional derivatives, line, surface and volume integrals, Stokes, Gauss and Green's theorems (without proofs) applications.

Ordinary Differential Equations: First order equation (linear and nonlinear), Second order linear differential equations with variable coefficients, Variation of parameters method, higher order linear differential equations with constant coefficients, Cauchy- Euler's equations, power series solutions, Legendre polynomials and Bessel's functions of the first kind and their properties.

Partial Differential Equations: Separation of variables method, Laplace equation, solutions of one dimensional heat and wave equations.

Probability and Statistics: Definitions of probability and simple theorems, conditional probability, Bayes Theorem, random variables, discrete and continuous distributions, Binomial, Poisson, and normal distributions, correlation and linear regression.

Numerical Methods: Solution of a system of linear equations by L-U decomposition, Gauss-Jordan and Gauss-Seidel Methods, Newton's interpolation formulae, Solution of a polynomial and a transcendental equation by Newton-Raphson method, numerical integration by trapezoidal rule, Simpson's rule and Gaussian quadrature, numerical solutions of first order differential equation by Euler's method and 4th order Runge-Kutta method.

SYLLABUS FOR M.PHARMACY**SECTION-A****General Aptitude (GA):**

(c) Verbal Ability: English grammar, sentence completion, verbal analogies, word groups, instructions, critical reasoning and verbal deduction.

(d) Numerical Ability: Numerical computation, numerical estimation, numerical reasoning and data interpretation.

SECTION-B**PHARMACEUTICS**

Micromeritics and Powder Rheology: Particle size and distribution, average particle size, number and weight distribution, particle number, methods for determining particle volume, methods of determining particle size- optical microscopy, sieving, sedimentation; measurements of particle shape, specific surface area; methods for determining surface area; permeability, adsorption, derived properties of powders, porosity, packing arrangement, densities, bulkiness & flow properties.

Viscosity and Rheology: Newtonian systems, Law of flow, kinematic viscosity, effect of temperature; non-Newtonian systems: pseudoplastic, dilatant, plastic; thixotropy, thixotropy in formulation, negative thixotropy, determination of viscosity, capillary, falling ball, rotational viscometers.

Dispersion Systems: Colloidal dispersions: Definition, types, properties of colloids, protective colloids, applications of colloids in pharmacy; Suspensions and Emulsions: Interfacial properties of suspended particles, settling in suspensions, theory of sedimentation, effect of Brownian motion, sedimentation of flocculated particles, sedimentation parameters, wetting of particles, controlled flocculation, flocculation in structured vehicles, rheological considerations; Emulsions-types, theories, physical stability. .

Liquid Dosages Forms: Introduction, types of additives used in formulations, vehicles, stabilizers, preservatives, suspending agents, emulsifying agents, solubilizers, colors, flavors and others, manufacturing packaging, labeling, evaluation of clear liquids, suspensions and emulsions official in pharmacopoeia

Semisolid Dosage Forms: Definitions, types, mechanisms of drug penetration, factor influencing penetration, semisolid bases and their selection. General formulation of semisolids, clear gels manufacturing procedure, evaluation and packaging;

Suppositories: Ideal requirements, bases, displacement value, manufacturing procedure, packaging and evaluation;

Pharmaceutical Aerosols: Definition, propellants, general formulation, manufacturing' and packaging methods, pharmaceutical applications;

Ophthalmic Preparations: Requirements, formulation, methods of preparation, labeling, containers, evaluation;

Capsules: Advantages and disadvantages of capsule dosage form, material for production of hard gelatin capsules, size of capsules, formulation, method of capsule filling, soft gelatin, capsule shell and capsule content, importance of base absorption and minimum/gm factors in soft capsules, quality control, stability testing and storage of capsule dosage forms.

Micro-encapsulation: Types of microcapsules, importance of microencapsulation in pharmacy, microencapsulation by phase separation, coacervation, multi-orifice, spray drying, spray congealing, polymerization complex emulsion, air suspension technique, coating pan and other techniques, evaluation of micro capsules.

Tablets: Advantages and disadvantages of tablets, Application of different types of tablets, Formulation of different types of tablets, granulation, technology on large-scale by various techniques, different types of tablet compression machinery and the equipments employed, evaluation of tablets. Coating of Tablets: Types of coating, film forming materials, formulation of coating solution, equipments for coating, coating process, evaluation of coated tablets. Stability kinetics and quality assurance.

Parenteral Products: Pre-formulation factors, routes of administration, water for injection, and sterile water for injection, pyrogenicity, non aqueous vehicles, isotonicity and methods of its adjustment, Formulation details, Containers and closures and selection, labeling; Pre-filling treatment, washing of containers and closures, preparation of solution and suspensions, filling and closing of ampoules, vials, infusion fluids, lyophilization& preparation of sterile powders, equipment for large scale manufacture and evaluation of parenteral products; Aseptic Techniques-source of contamination and methods of prevention, Design of aseptic area, Laminar flow bench services and maintenance. Sterility testing of pharmaceuticals.

Packaging of Pharmaceutical Products: Packaging components, types, specifications and methods of evaluation, stability aspects of packaging. Packaging equipments, factors influence choice of containers, legal and official requirements for containers, package testing.

Designing of dosage forms: Pre-formulation studies, Study of physical properties of drug like physical form, particle size, shape, density, wetting, dielectric constant. Solubility, dissolution and organoleptic properties and their effect on formulation, stability and bioavailability. Study of chemical properties of drugs like hydrolysis, oxidation, reduction, racemization, polymerization etc., and their influence on formulation and stability of products. Study of pro-drugs in solving problems related to stability, bioavailability and elegance of formulations. Design, development and process validation methods for pharmaceutical operations involved in the production of pharmaceutical products with special reference to tablets, suspensions. Stabilization and stability testing protocol for various pharmaceutical products. ICH guidelines for stability testing of formulations.

Performance evaluation methods: In-vitro dissolution studies for solid dosage forms methods, interpretation of dissolution data. Bioavailability studies and bioavailability testing protocol and procedures. In vivo methods of evaluation and statistical treatment. GMP and quality assurance, Quality audit. Design, development, production and evaluation of controlled/sustained/extended release formulations.

Biopharmaceutic: Passage of drugs across biological barrier (passive diffusion, active transport, facilitated diffusion, ion-pair formation and pinocytosis); Factors influencing absorption- biological, physico-chemical, physiological and pharmaceutical; Drug distribution in the body, plasma protein binding.

Pharmacokinetics: Significance of plasma drug concentration measurement. Compartment model- Definition and Scope. Pharmacokinetics of drug absorption - Zero order and first order absorption rate constant using Wagner-Nelson and residual methods. Volume of distribution and distribution coefficient. Compartment kinetics- One compartment and two compartment models. Determination of pharmacokinetic parameters from plasma and urine data after drug administration by intravascular and oral route. Clearance concept, mechanism of renal clearance, clearance ratio, determination of renal clearance. Extraction ratio, hepatic clearance, biliary excretion, extra-hepatic circulation. Non-linear pharmacokinetics with special reference to one compartment model after I.V. drug administration.

Clinical Pharmacokinetics: Definition and scope: Dosage adjustment in patients with and without renal and hepatic failure; Design of single dose bio-equivalence study and relevant statistics; Pharmacokinetic drug interactions and their significance in combination therapy.

Bioavailability and bioequivalence: Measures of bioavailability, C_{max} , t_{max} , K_{el} and Area Under the Curve (AUC); Design of single dose bioequivalence study and relevant statistics; Review of regulatory requirements for conducting bioequivalent studies. Biopharmaceutical Classification System (BCS) of drugs.

PHARMACEUTICAL CHEMISTRY

Basic Principles: Physico-chemical and stereoisomeric (Optical, geometrical) aspects of drug molecules and biological action, Bioisosterism, Drug-receptor interactions including transduction mechanisms; Drug metabolism and Concept of Prodrugs; Principles of Drug Design (Theoretical Aspects) Traditional analog and mechanism based approaches, QSAR approaches, Applications of quantum mechanics, Computer Aided Drug Designing (CADD) and molecular modelling

Synthetic Procedures, Mode of Action, Uses, Structure Activity Relationships including physicochemical Properties of the Following Classes of Drugs

Drugs acting at synaptic and neuro-effector junction sites; Cholinergics, anti-cholinergics and cholinesterase inhibitor; Adrenergic drugs; Antispasmodic and anti-ulcer drugs; Local Anesthetics; Neuromuscular blocking agents; Autacoids, Steroidal Drugs; Drugs acting on the central nervous system; Diuretics; Anti-hypertensives; Anti-arrythmic agents, anti-anginal agents, Cardiotonics; Anti-hyperlipedemic agents; Anticoagulants and Anti-platelet drugs; Thyroid and Anti thyroid drugs; Insulin and oral hypoglycemic agents; Chemotherapeutic Agents used in bacterial, fungal, viral, protozoal, parasitic and other infections, Antibiotics: β -Lactam, macrolides, tetracyclines, aminoglycosides, polypeptide antibiotics, fluoroquinolones; Anti-neoplastic agents; Anti-viral agents (including anti-HIV); Immunosuppressives and immunostimulants; Diagnostic agents; Pharmaceutical Aids;

Basic Principles of pharmaceutical analysis: Preliminaries and definitions, Significant figures, Rules for retaining significant digits, Types of errors, Mean deviation, Standard deviation, Statistical treatment of small data sets, Selection of sample, Precision and accuracy,

Chromatography: Theory of chromatography, plate theory, Factors affecting resolution, van Deemter equation, The following chromatographic techniques (including instrumentation) with relevant examples of Pharmacopoeial products: TLC, HPLC, GLC, HPTLC, Paper Chromatography and Column Chromatography;

Theoretical Aspects, Basic Instrumentation, Elements of Interpretation of Spectra, and applications (quantitative and qualitative) of the following

Ultraviolet and visible spectrophotometry, Fluorimetry, Infrared spectrophotometry, Nuclear Magnetic Resonance spectroscopy, Mass Spectrometry (EI & CI only), Flame Photometry, Atomic Absorption Spectroscopy, X-ray Diffraction Analysis, Radioimmunoassay.

PHARMACOLOGY

Basic Principles of Cell Injury and Adaptations: Causes of Cellular injury, pathogenesis, morphology of cell injury, adaptations and cell death.

Basic Mechanisms involved in the process of inflammation and repair: Vascular and cellular events of acute inflammation, chemical mediators of inflammation, pathogenesis of chronic inflammation, brief outline of the process of repair.

Immunopathophysiology: T and B cells, MHC proteins, antigen presenting cells, immune tolerance, pathogenesis of hypersensitivity reactions, autoimmune diseases, AIDS, Amyloidosis.

Pathophysiology of Common Diseases: Asthma, diabetes, rheumatoid arthritis, gout, ulcerative colitis, neoplasia, psychosis, depression, mania, epilepsy, acute and chronic renal failure, hypertension, angina, congestive heart failure, atherosclerosis, myocardial infarction, congestive heart failure, peptic ulcer, anemias, hepatic disorders, tuberculosis, urinary tract infections and sexually transmitted diseases. Wherever applicable the molecular basis should be discussed.

Fundamentals of general pharmacology: Dosage forms and routes of administration, mechanism of action, combined effect of drugs, factors modifying drug action, tolerance and dependence; Pharmacogenetics; Principles of Basic and Clinical pharmacokinetics, absorption, Distribution, Metabolism and Excretion of drugs, Adverse Drug Reactions; Bioassay of Drugs and Biological Standardization; Discovery and development of new drugs, Bioavailability and bioequivalence studies;

Pharmacology of Peripheral Nervous System: Neurohumoral transmission (autonomic and somatic), Parasympathomimetics, Parasympatholytics, Sympathomimetics, Adrenergic receptor and neuron blocking agents, Ganglion stimulants and blocking agents, Neuromuscular blocking Agents, Local anesthetic Agents.

Pharmacology of Central Nervous System: Neurohumoral transmission in the C.N.S., General Anesthetics, Alcohols and disulfiram, Sedatives, Hypnotics, Anti-anxiety agents and Centrally acting muscle relaxants, Psychopharmacological agents (anti-psychotics), anti-manics and hallucinogens, Antidepressants, Anti-epileptics drugs, Anti-Parkinsonian drugs, Analgesics, Antipyretics, Narcotic analgesics and antagonists, C.N.S. stimulants, Drug Addiction and Drug Abuse.

Pharmacology of Cardiovascular System: Drugs used in the management of congestive cardiac failure, Antihypertensive drugs, Anti-anginal and Vasodilator drugs, including calcium channel blockers and beta adrenergic antagonists, Anti-arrhythmic drugs, Anti-hyperlipidemic drugs, Drugs used in the therapy of shock.

Drugs Acting on the Hemopoietic System: Hematinics, Anticoagulants, Vitamin K and hemostatic agents, Fibrinolytic and anti-platelet drugs, Blood and plasma volume expanders.

Drugs acting on urinary system: Fluid and electrolyte balance, Diuretics.

Autacoids: Histamine, Antihistaminic drugs, 5-HT- its agonists and antagonists, Prostaglandins, thromboxanes and leukotrienes, Angiotensin, Bradykinin and Substance P and other vasoactive peptides, non-steroidal anti-inflammatory and anti-gout agents.

Drugs Acting on the Respiratory System: Anti-asthmatic drugs including bronchodilators, Anti-tussives and expectorants, Respiratory stimulants.

Drugs acting on the Gastrointestinal Tract: Antacids, Anti-secretory and Anti-ulcer drugs, Laxatives and anti-diarrhoeal drugs, Appetite Stimulants and Suppressants, Emetics and anti-emetics, Miscellaneous: Carminatives, demulcents, protectives, adsorbents, astringents, digestants, enzymes and mucolytics.

Pharmacology of Endocrine System: Hypothalamic and pituitary hormones, Thyroid hormones and anti thyroid drugs, parathormone, calcitonin and Vitamin D, Insulin, glucagons, incretins, oral hypoglycemic agents and insulin analogs, ACTH and corticosteroids, Androgens and anabolic steroids, Estrogens, progesterone and oral contraceptives, Drugs acting on the uterus.

Chemotherapy: General Principles of Chemotherapy, Bacterial resistance; Sulfonamides and cotrimoxazole, Antibiotics- Penicillins, Cephalosporins, Aminoglycosides, Chloramphenicol, Macrolides, Tetracyclines, Quinolones, fluoroquinolones and Miscellaneous antibiotics; Chemotherapy of tuberculosis, leprosy, fungal diseases, viral diseases, HIV and AIDS, urinary tract infections and sexually transmitted diseases, malaria, amoebiasis and other protozoal infections and Anthelmintics. Chemotherapy of malignancy and immunosuppressive agents.

Principles of Toxicology: Definition of poison, general principles of treatment of poisoning with particular reference to barbiturates, opioids, organophosphorous and atropine poisoning, Heavy metals and heavy metal antagonists.

Basic Concepts of Pharmacotherapy: Clinical Pharmacokinetics and individualization of Drug therapy, Drug delivery systems and their Biopharmaceutic & Therapeutic considerations, Drugs used during infancy and in the elderly persons (Pediatrics & Geriatrics), Drugs used during pregnancy, Drug induced diseases, The basics of drug interactions, General principles of clinical toxicology, Common clinical laboratory tests and their interpretation

Important Disorders of Organs, Systems and their Management: Cardio-vascular disorders- Hypertension, Congestive heart failure, Angina, Acute myocardial infarction, Cardiac arrhythmias; CNS Disorders: Epilepsy, Parkinsonism, Schizophrenia; Depression Respiratory disease-Asthma; Gastrointestinal Disorders-Peptic ulcer, Ulcerative colitis, Hepatitis, Cirrhosis; Endocrine Disorders- Diabetes mellitus and Thyroid disorders; Infectious Diseases-Tuberculosis, Urinary tract infections, Enteric infections, Upper respiratory infections; Hematopoietic Disorders- Anemias; Joint and Connective tissue disorders-Rheumatic diseases, Gout and Hyperuricemia; Neoplastic Diseases-

Acute Leukaemias, Hodgkin's disease. Therapeutic Drug Monitoring, Concept of Essential Drugs and Rational Drug use.

PHARMACOGNOSY

Sources of Drugs: Biological, marine, mineral and plant tissue cultures as sources of drugs

Classification of Drugs: Morphological, taxonomical, chemical and pharmacological classification of drugs

Study of medicinally important plants belonging to the families with special reference to: Apocynaceae, Solanaceae, Rutaceae, Umbelliferae, Leguminosae, Rubiaceae, Liliaceae, Graminae, Labiatae, Cruciferae, Papaveraceae;

Quality Control of Crude Drugs: Adulteration of crude drugs and their detection by organoleptic, microscopic, physical, chemical and biological methods and properties.

Introduction to Active Constituents of Drugs: their isolation, classification and properties.

Phytochemical Screening: Preparation of extracts, Screening of alkaloids, saponins, cardenolides and bufadienolides, flavonoids and leucoanthocyanidins, tannins and polyphenols, anthraquinones, cynogenetic glycosides, amino acids in plant extracts;

Systematic pharmacognostic study of the followings:

Carbohydrates and derived products: Agar, Guar gum, Acacia, Honey, Isabgol, Pectin, Starch, Sterculia and Tragacanth; Lipids: Bees wax, Castor oil, Cocoa butter, Codliver oil, Hydnocarpus oil, Kokum butter, Lard, Linseed oil, Rice, Bran oil, Shark liver oil and Wool fat; Resins: Study of Drugs Containing Resins and Resin Combinations like Colophony, podophyllum, jalap, cannabis, capsicum, myrrh, asafoetida, balsam of Tolu, balsam of Peru, benzoin, turmeric, ginger; Tannins: Study of tannins and tannin containing drugs like Gambier, black catechu, gall and myrobalan; Volatile Oils: General methods of obtaining volatile oils from plants, Study of volatile oils of Mentha, Coriander, Cinnamon, Cassia, Lemon peel, Orange peel, Lemon grass, Citronella, Caraway, Dill, Spearmint, Clove, Fennel, Nutmeg, Eucalyptus, Chenopodium, Cardamom, Valerian, Musk, Palmarosa, Gaultheria, Sandal wood;

Fibres: Study of fibers used in pharmacy such as cotton, silk, wool, nylon, glass-wool, polyester and asbestos.

Study of the biological sources, cultivation, collection, commercial varieties, chemical constituents, substitutes, adulterants, uses, diagnostic macroscopic and microscopic features and specific chemical tests of following groups of drugs:

Glycoside Containing Drugs: saponins :iquorice, ginseng, dioscorea, sarsaparilla, and senega; Cardioactiveglycosides:digitalis, squill, strophanthus and thevetia; Anthraquinonecathartics:aloe, senna, rhubarb and cascara; Others:Psoralea, Ammimajus, Ammivisnaga, gentian, saffron, chirata, quassia.

Alkaloid Containing Drugs: Pyridine-piperidine:tobacco, areca and lobelia; Tropane: belladonna, hyoscyamus, datura, duboisia, coca and withania;Quinoline and Isoquinoline:Cinchona, ipecac, opium; Indole:ergot, rauwolfia, catharanthus, nux-vomica and physostigma;Imidazole:pilocarpus; Steroidal:veratrum and kurch; AlkaloidalAmine:ephedra and colchicum;Glycoalkaloid:solanum; Purines:coffee, tea and cola.

Biological sources, preparation, identification tests and uses of the following enzymes: Diastase, papain, pepsin, trypsin, pancreatin.

SECTION-C

BIOCHEMISTRY

Biochemistry in pharmaceutical sciences: The concept of free energy, Determination of change in free energy - from equilibrium constant and reduction potential, bioenergetics, production of ATP and its biological significance;

Enzymes: Nomenclature, enzyme kinetics and their mechanism of action, mechanism of inhibition, enzymes and iso-enzymes in clinical diagnosis;

Co-enzymes: Vitamins as co-enzymes and their significance. Metals as cofactors and their significance; Carbohydrate Metabolism: Conversion of polysaccharides to glucose-1-phosphate, Glycolysis, fermentation and their regulation, Gluconeogenesis and glycogenolysis, Metabolism of galactose and galactosemia, Role of sugar nucleotides in biosynthesis, and Pentose phosphate pathway;

The Citric Acid Cycle: Significance, reactions and energetics of the cycle, Amphibolic role of the cycle, and Glyoxalic acid cycle;

Lipids Metabolism :Oxidation of fatty acids, β -oxidation & energetics, biosynthesis of ketone bodies and their utilization, biosynthesis of saturated and unsaturated fatty acids, Control of lipid metabolism, Essential fatty acids & eicosanoids (prostaglandins, thromboxanes and leukotrienes), phospholipids, and sphingolipids, Biosynthesis of eicosanoids, cholesterol, androgens, progesterone, estrogens corticosteroids and bile acids;

Biological Oxidation: Redox-potential, enzymes and co-enzymes involved in oxidation reduction & its control, The respiratory chain, its role in energy capture and its control, energetics of oxidative

phosphorylation. Inhibitors of respiratory chain and oxidative phosphorylation, Mechanism of oxidative phosphorylation;

Metabolism of ammonia and nitrogen containing monomers: Nitrogen balance, Biosynthesis of amino acids, Catabolism of amino acids, Conversion of amino acids to specialized products, Assimilation of ammonia, Urea. cycle, metabolic disorders of urea cycle, Metabolism of sulphur containing amino acids;

Purine biosynthesis: Purine nucleotide inter-conversions;

Pyrimidine biosynthesis and formation of deoxyribonucleotides;

Biosynthesis of Nucleic Acids: Brief introduction of genetic organization of the mammalian genome, alteration and rearrangements of genetic material, Biosynthesis of DNA and its replications;

Mutation: Physical & chemical mutagenesis/carcinogenesis, DNA repair mechanism. Biosynthesis of RNA;

Genetic Code and Protein Synthesis: Genetic code, Components of protein synthesis and Inhibition of protein synthesis.

MICROBIOLOGY

Structure and Classification of microbes and their taxonomy: Actinomycetes, bacteria, rickettsiae, spirochetes and viruses;

Identification of Microbes: Stains and types of staining techniques, electron microscopy; Nutrition, cultivation, isolation of bacteria, actinomycetes, fungi, viruses, etc; Microbial genetics and variation;

Control of microbes by physical and chemical methods: Disinfection, factors influencing disinfectants, dynamics of disinfection, disinfectants and antiseptics and their evaluation;

Sterilization: different methods, validation of sterilization methods & equipments; Sterility testing of all pharmaceutical products. Microbial assays of antibiotics, vitamins & amino acids.

Immunology and Immunological Preparations: antigens and heptans, immune system, cellular/humoral immunity, immunological tolerance, antigen-antibody reactions and their applications. Hypersensitivity, active and passive immunization. Vaccines and sera: their preparation, standardization and storage.

Genetic Recombination: Transformation, conjugation, transduction, protoplast fusion and gene cloning and their applications. Development of hybridoma for monoclonal antibodies. Study of drugs produced by biotechnology such as Activase, Humulin, Humatrope, HB etc;

Antibiotics: Historical development of antibiotics. Antimicrobial spectrum and methods used for their standardization. Screening of soil for organisms producing antibiotics, fermenter, its design,

control of different parameters. Isolation of mutants, factors influencing rate of mutation. Design of fermentation process. Isolation of fermentation products with special reference to penicillins, streptomycin, tetracyclines and vitamin B12.

PHARMACEUTICAL JURISPRUDENCE (C)

Pharmaceutical Legislations: A brief review of Drugs & Pharmaceutical Industry; Pharmaceutical Education-

An elaborate study of the followings: Pharmaceutical Ethics; Pharmacy Act 1948; Drugs and Cosmetics Act 1940 and Rules 1945; Medicinal & Toilet Preparations (Excise Duties) Act 1955; Narcotic Drugs & Psychotropic Substances Act 1985 & Rules; Drugs Price Control Order;

A brief study of the following Acts with special reference to the main provisions and the latest amendments: Poisons Act 1919; Drugs and Magic Remedies (Objectionable Advertisements) Act 1954; Medical Termination of Pregnancy Act 1970 & Rules 1975; Prevention of Cruelty to Animals Act 1960; States Shops & Establishments Act & Rules; Insecticides Act 1968; AICTE Act 1987; Factories Act 1948; Minimum Wages Act 1948; Patents Act 1970.

A brief study of the various Prescription/Non-prescription Products. Medical/Surgical accessories, diagnostic aids, appliances available in the market.

Syllabus for MBA & MCA

Section-A

Verbal Ability:

Vocabulary Based (Synonyms Antonyms), English Usage or Grammar, Sentence Correction, Fill in the blanks, Close Passage, Analogies or Reverse Analogies, Jumbled Paragraph, Meaning-Usage Match, Summary Questions, Verbal Reasoning, Facts/Inferences/Judgements, Reading Comprehension

Section-B

Quantitative Ability:

Geometry (Lines, angles, Triangles, Spheres, Rectangles, Cube, Cone etc.), Ratios and Proportion, Percentages, In-equations Quadratic and linear equations, Algebra, Profit & Loss Averages, Partnership (Accounts), Time-Speed-Distance, Work and time, Number system, HCF & LCM Geometric Progression, Arithmetic progression, Arithmetic mean, Geometric mean, Harmonic mean, Median, Mode, Number Base System, BODMAS, Menstruation, Allegation & Mixtures, Work, Pipes and Cisterns, Simple Interest & Compound Interest, Set Theory, Venn diagram, Installment Payments, Clocks, Probability, Permutations Combinations, Trigonometry, Vectors, Binomial Expansion, Co-ordinate geometry, Logarithm, Calendar, Maxima & Minima Progression, Surds & Indices, Complex numbers

Section-C

Data Interpretation and Reasoning:

There will be questions of data interpretation which will be mostly based of various graphs.

Graphs

Column graphs, Bar Graphs, Line charts, Pie Chart, Graphs representing Area, Venn diagram etc.

Reasoning

Critical reasoning, Visual reasoning, Assumption-Premise-Conclusion, Assertion and reasons, Statements and assumptions, Identifying valid inferences, Identifying arguments, Statements and conclusions, Cause and Effect, Identifying probably true probably false, Definitely true definitely false kind of statement, Linear arrangements, Matrix arrangements (Puzzles,

Syllogisms, Functions), Family tree - identifying relationship among group of people, Symbol Based problems, Coding and decoding, Sequencing etc.

Section-D

Business Awareness (Only for MBA)

Current Affairs, General Knowledge, Business, Punch line of companies, Top officials of big companies, Major corporate events, Famous award and prizes, World Records, Books and authors, Science, History, Geography, International organizations, Important quotations, Social issues, Sports, Finance, Automobiles, Entertainment, Politics etc.

Computer Awareness (Only for MCA)

Current Affairs, General Knowledge, Elements of computers, Number systems, Basic electronic gates, Boolean algebra, Flip-Flops. Systems, Programming and Operating System, Fundamentals of 'C' language

Detail of Intake in Campus of Himachal Pradesh Technical University

S. No.	Name of Principal/College Address and Website	Civil Engineering	Information Technology	Computer Science & Engg.	Electronics & Comm. Engg.	Electrical Engineering	Mechanical Engineering	MBA	MCA
1	Campus, Himachal Pradesh Technical University,, Hamirpur, District Hamirpur-177001 (HP) Website:- www.himtu.ac.in Phone No:01972-224152, 224153, 224180	-	-	-	-	-	-	40	40
2	Himachal Pradesh Technical University Rajiv Gandhi Government Engineering College Campus NagrotaBagwan, District Kangra-176047 (H.P.) Phone No:01892-249188	-	-	-	-	-	-	60	-

Seats Available in Different Institutions along with courses and their intake/(Actual Admission) affiliated to Himachal Pradesh Technical University for 2015-16 in Under Graduate Programmes which shall normally be followed for 2016-17 also.

S. No.	Name of Principal/College Address and Website	Civil Engineering	Information Technology	Computer Science & Engg.	Electronics & Comm. Engg.	Electrical Engineering	Mechanical Engineering	Electrical & Electronics Engg.	Automobile Engg.	Textile	B.Pharmacy	B.Pharmacy (Ayurveda)
1	J.N. Govt. Engineering College, Bhechandhar, Sundernagar, Distt. Mandi-175018 Phone No. 01907-267199, 267688, 266728 & 267504 (Telefax) Website: www.ingec.ac.in	60	0	0	60	0	60	0	0	60	-	-
2	AtalBihari Vajpayee Govt Institute of Engineering and Technology, Pragatinagar, Distt. Shimla-171202 (HP) Phone No : 01783-253918, Website: www.abvgiet.ac.in	0	0	60	60	0	0	0	0	0	-	-
3	College of Ayurveda Pharmaceutical Sciences, Jogindernagar, Distt. Mandi-175015 (HP) Phone No: 01908-222931(telefax) Website: www.jogindernagar.com	-	-	-	-	-	-	-	-	-	-	30
4	Government College of Pharmacy, Rohru, Distt. Shimla-171207 (HP) Phone No: 01781-241306 Website: www.bpharmacyrohru.org	-	-	-	-	-	-	-	-	-	40	-
5	Rajiv Gandhi Government Engineering College Campus NagrotaBagwan, District Kangra-176047 (H.P.) Phone No:01892-249188	60	-	-	-	-	60	-	-	-	-	-
6	Govt. Engg. College Jeory, Rampur, Distt. Shimla. Camp office at JNGEC Sundernagar, Distt.Mandi (H.P.)-174401	60	-	-	60	-	60	-	-	-	-	-
7	Bells Institute of Management & Technology, Knowledge City Mehli, Distt. Shimla-171013(H.P.) Phone No. 01772-620213, 2620313, Website www.bells.ac.in	90	0	90	60	60	90	0	0	-	-	-
8	DevBhumy Institute of Engineering & Technology, Village Chandpur, Tehsil Haroli, Distt. Una-177220(H.P.) Phone No. 01975-205594 (Telefax), Website : www.dbgi.org.in	60	0	60	60	0	90	0	60	-	-	-
9	Green Hills Engineering College, Gandhi Gram, KumarhattiNahan Road, P.O. Bohali, Distt. Solan-173229(H.P.) Phone No. 01792-266772, 266773, 266086 (fax) Website : www.ghcc.co.in	120	60	60	90	60	120	0	0	-	-	-

10	Himachal Institute of Engineering & Technology, Vidyanagar, Near Central University, P.O. Dramman, Tehsil Shahpur, Distt. Kangra-176223 Phone No. 01892-398100, 398150 (fax), Website : www.hiet.co.in	90	0	60	60	60	120	60	0	-	-	-
11	Himachal Institute of Technology, Rampur Ghat Road, Opposite Dental College Paonta Sahib, Distt. Sirmour-173025(H.P.) Phone No. 01704-223726 & 223298, Website : www.hids.ac.in	40	0	60	40	40	40	0	0	-	-	-
12	Himalayan Institute of Engineering & Technology, Kala Amb, Tehsil Nahan, Distt. Sirmour-173030 (H.P.) Phone No. 01702-308500 to 308508, 308502(fax) Website: www.himalayaninstitution.com	120	0	120	120	60	120	0	0	-	-	-
13	IITT College of Engineering, Kala Amb, Village JohronTrilokpur Road, Kala Amb, Distt. Sirmour-173030(H.P.)Phone No. 01702-238915 & 238226 (fax)Website: www.iitt.india.com	0	60	120	120	60	120	0	0	-	-	-
14	K.C. Group of Research and Professional Institute, Pandoga, V.P.O. UparlaPandoga, Tehsil Haroli, Distt. Una-177207(H.P.) Phone No. 01975-241461, 241111, Website: www.kcinstitutes.com	90	0	60	60	60	120	0	0	-	-	-
15	L.R. Institute of Engineering & Technology, Village Jabli-Kyar, P.O. Oachghat, Sultanpur Road, Distt. Solan-173223 (H.P.) Phone No. 01792-252857 (Telefax), 304804 & 304800, Website : www.lriet.org	90	60	90	90	60	90	0	0	-	-	-
16	MIT College of Engineering & Management, Village GanohRajputan, P.O. Bani, Tehsil Barsar, Distt. Hamirpur-174304(H.P.) Phone No. 01972-302200, 302214 (fax), 215234, Website : www.mithmr.com	120	0	60	60	60	120	60	0	-	-	-
17	Shiva Institute of Engineering & Technology, Village LuhanooKanaitan, P.O. Chandpur, Tehsil Sadar, Distt. Bilaspur-174004(H.P.) Phone No. 01978- 304001-304004 & 304005 (fax), Website : www.sieth.ac.in	120	0	60	40	0	120	60	0	-	-	-
18	Shiva Educational Society Abdhanighat (Panol), Tehsil Ghumarwin, Distt. Bilaspur (H.P.) – 174021	-	-	-	-	-	-	-	-	-	60	-
19	SIRDA Institute of Engineering Technology NH-21, Naulakha, P.O Kanaid, Tehsil Sundernagar, Distt. Mandi-175019(H.P.) Phone No. 01907-262612, 262512 & 262312 (fax) Website : www.sirda.org	120	60	90	90	0	120	120	0	-	-	-
20	T.R. Abhilashi Memorial Institute of Engineering & Technology, Village Tanda, P.O. Balt (Ner-Chowk), Tehsil Sadar, Distt. Mandi-175008(H.P.) Phone No. 01905-24819, & 248189 (fax), Website : www.tramiet.in	120	0	60	60	60	120	0	0	-	-	-

21	Vaishno College of Engineering, Village Thapkur, P.O. Bhadroya, Tehsil Nurpur, Distt. Kangra-176403(H.P.) Phone No.01893-233444,233400 (fax), Website : www.vcoe.co.in	120	0	60	60	60	90	0	0	-	-	-
22	Abhilashi College of Pharmacy Village Tanda, P.O Balt (Ner-Chowk) Tehsil Sadar, Distt. Mandi-175008 (HP) Phone No: 01905-248539 Website: www.abhilashi-edu.org	-	-	-	-	-	-	-	-	-	120	-
23	DDM College of Pharmacy, GondpurBanehra (Upper), Tehsil Amb, Distt. Una-177312 Phone No: 01976-268868, 268568 Website: www.ddminstitute.com	-	-	-	-	-	-	-	-	-	60	-
24	Dreamz College of Pharmacy, Village Khilra, P.O. Meramasit, Tehsil Sundernagar, Distt. Mandi-175036 Phone No: 01905-248598, 248599 Website: www.dreamzedu.in	-	-	-	-	-	-	-	-	-	120	-
25	Himachal Institute of Pharmaceutical Education & Research (HIPER), VPO Nadaun, Distt. Hamirpur-177033 Phone No: 01972-201113, 232893 Website: www.hiper.co.in	-	-	-	-	-	-	-	-	-	60	-
26	Himachal Institute of Pharmacy, Rampur Ghat Road, Paonta Sahib, Distt. Sirmour-173025 Phone No: 01704-223298, 223039 Website: www.hids.in	-	-	-	-	-	-	-	-	-	60	-
27	Himalayan Institute of Pharmacy, Kala-Amb, Tehsil Nahan, Distt. Sirmour-173030 (HP) Phone No: 01702-308500 to 308508 Website: www.himalayaninstitutions.com	-	-	-	-	-	-	-	-	-	60	-
28	K.C. Institute of Pharmaceutical Sciences, V.P.O. PandogaUparla, Distt. Una-177207 Phone No. 01975-241461 Website: www.kcinstitutes.in	-	-	-	-	-	-	-	-	-	120	-
29	L.R. Institute of Pharmacy, Village Jabli-Kyar, P.O. Oachghat, Sultanpur Road, Distt. Solan-173223 (HP) Phone No: 01792-252854, 252857 Website: www.lrinstitute.org .	-	-	-	-	-	-	-	-	-	60	-
30	Laureate Institute of Pharmacy, VPO Kathog, Tehsil Jawalaji, Distt. Kangra-177101 Phone No: 01970-222847(HP) Website: www.laureateinstitute.in	-	-	-	-	-	-	-	-	-	120	-
31	Sai Ram Education Trust's College of B.Pharmacy VPO Tipper, Tehsil Barsar, Distt. Hamirpur-174312 Phone No: 01972-285696,97 Website: www.sairamedutech.org	-	-	-	-	-	-	-	-	-	40	-
32	Vinayaka College of Pharmacy, Village Bahoguna, P.O. Garsa, Tehsil Bhuntar, Distt. Kullu-175141 Phone No: 01902-272277 Website: www.vinayakainstitutes.org	-	-	-	-	-	-	-	-	-	60	-
33	Himachal Pharmacy College, Village Maganpura, P.O. Manjauli, Tehsil Nalagarh, Distt. Solan (H.P.) – 174101	-	-	-	-	-	-	-	-	-	60	-

Seats Available in Different Institutions affiliated to Himachal Pradesh Technical University in Post Graduate Programmes

S. No.	Name of Principal/College Address and Website	Seats Available in Different Programmes									
		M.Tech					MBA	MCA	M. Pharma		
		Civil Engineering	Computer Science & Engineering	Mechanical Engineering	Electrical Engineering	Electronics & Communication Engineering			Pharmaceutics	Pharmaceutical chemistry	Pharmaceutical Analysis and Quality Assurance
1	Govt. P. G. College Dharamashala, Distt. Kangra H.P.) – 176215	-	-	-	-	-	60	-	-	-	-
2	KLBDVAV College for Girls Management Palampur, Kangra (H.P.) – 176061	-	-	-	-	-	60	-	-	-	-
3	K. C. Educational and welfare society, Pandoga, Una	-	-	-	-	-	120	-	-	-	-
4	Gautamgirls College, Hamirpur (H.P.) – 177001	-	-	-	-	-	120	-	-	-	-
5	Green Hills Engineering College, Gandhi Gram, KumarhattiNahan Road, P.O. Bohali, Distt. Solan-173229(H.P.) Phone No. 01792-266772, 266773, 266086 (fax) Website : www.ghcec.co.in	0	0	18	0	18	-	-	-	-	-
6	L.R. Institute of Engineering & Technology, Village Jabli-Kyar, P.O. Oachghat, Sultanpur Road, Distt. Solan-173223 (H.P.) Phone No. 01792-252857 (Telefax), Website : www.lriet.org	18	18	18	0	24	-	-	-	-	-
7	IITT College of Engineering, Kala Amb, Village JohronTrilokpur Road, Kala Amb, Distt. Sirmour-173030(H.P.)Phone No. 01702-238915 & 238226 (fax) Website: www.iitt.india.com	0	0	24	18	0	-	-	-	-	-
8	Himalayan Institute of Engineering & Technology, Kala Amb, Tehsil Nahan, Distt. Sirmour-173030 (H.P.) Phone No. 01702-308500 to 308508,308502 (fax) Website: www.himalayaninstitution.com	0	24	0	0	24	-	-	-	-	-
9	SIRDA Institute of Engineering Technology NH-21, Naulakha, P.O Kanaid, Tehsil Sundernagar, Distt. Mandi-175019(H.P.) Phone No. 01907-262612, 262512 & 262312 (fax) Website : www.sirda.org	24	24	0	0	0	-	-	-	-	-
10	Bells Institute of Management & Technology, Knowledge City Mehli, Distt. Shimla-171013(H.P.) Phone No. 01772-620213, 2620313, Website www.bells.ac.in	0	24	0	0	0	-	-	-	-	-

11	Vaishno College of Engineering, Village Thapkour, P.O. Bhadroya, Tehsil Nurpur, Distt. Kangra-176403(H.P.) Phone No.01893-233444,233400 (fax), Website : www.vcoe.co.in	0	24	0	0	24	-	-	-	-	-
12	Shiva Institute of Engineering & Technology LuhnooKanatain, P.O. Chandpur, Tehsil & Distt. Bilaspur (H.P.) Phone No. 01978-304001-304004 & 304005 (fax) Website : www.sieth.ac.in	0	18	0	0	0	-	-	-	-	-
13	Govt. Post Graduate College, Dharamshala-176215 Phone No: 01892-224894 Website www.gcdharamshala.ac.in	-	-	-	-	-	60	-	-	-	-
14	Govt. College Una, Nangal Road, Una-174303 Phone No:01975-233423 Website: www.govtpgcollegeuna.in	-	-	-	-	-	60	-	-	-	-
15	Gautam Institute of Management & Technology, Ward No. 10, Hamirpur-177001 Phone No: 01972-221493 Website: www.gautamcollege.com	-	-	-	-	-	120	-	-	-	-
16	K.L.B. D.A.V. Girls College for Management, Lohna, Tehsil Palampur, Distt. Kangra-176061 (HP) Phone No: 01894-232587 Website: www.klvdavcollege.org	-	-	-	-	-	60	-	-	-	-
17	Abhilashi Institute of Management Studies, NerChowk, Tehsil Sadar, Distt. Mandi-175008 Phone No: 01905-243328 Website: www.abhilashi-edu.in	-	-	-	-	-	60	-	-	-	-
18	Bells Institute of Management & Technology, Knowledge City Mehli, Distt. Shimla-171013(H.P.) Phone No. 01772-620213, 2620313, Website www.bells.ac.in	-	-	-	-	-	180	-	-	-	-
19	Himalayan Institute of Management, Sadhora Road, Kala-Amb, Tehsil Nahan, Distt. Sirmour-173030 Phone No: 01702-308500 Website: www.himalayaninstitution.com	-	-	-	-	-	120	-	-	-	-
20	H.P. Institute of Management Studies, Airport Road, Totu, Shimla-171011 Phone No: 0177-2837711 Website: www.hpcmanagement.org	-	-	-	-	-	120	-	-	-	-
21	K.C. Group of Research & Professional Institute, V.P.O. UparlaPandoga, Tehsil Haroli, Distt. Una-177207 Phone No: 01975-241461 Website: www.kcinstitute.com	-	-	-	-	-	180	-	-	-	-
22	L.R. Institute of Management, Village Jabli-Kyar, P.O. Oachghat, Tehsil Solan-173223 Phone No: 01792-252524 Website : www.lrinstitute.com	-	-	-	-	-	120	-	-	-	-

23	Himalayan Institute of Pharmacy, Kala-Amb, Tehsil Nahan, Distt. Sirmour-173030 (HP) Phone No: 01702-308500 to 308508 Website: www.himalayaninstitutions.com	-	-	-	-	-	-	-	24	-	-
24	Abhilashi College of Pharmacy Village Tanda, P.O Balt (Ner-Chowk) Tehsil Sadar, Distt. Mandi-175008 (HP) Phone No: 01905-248539 Website: www.abhilashi-edu.org	-	-	-	-	-	-	-	-	-	18
25	Himachal Institute of Pharmacy, Rampur Ghat Road, Paonta Sahib, Distt. Sirmour-173025 Phone No: 01704-223298, 223039 Website: www.hids.in	-	-	-	-	-	-	-	24	-	-
26	L.R. Institute of Pharmacy, Village Jabli-Kyar, P.O. Oachghat, Sultanpur Road, Distt. Solan-173223 (HP) Phone No: 01792-252854, 252857 Website: www.lrinstitute.org .	-	-	-	-	-	-	-	18	-	-
27	Laureate Institute of Pharmacy, VPO Kathog, Tehsil Jwalaji, Distt. Kangra-177101 Phone No: 01970-222847(HP) Website: www.laureateinstitute.in	-	-	-	-	-	-	-	18	-	18
28	Govt. College Una, Nangal Road, Una-174303 Phone No:01975-233423 Website: www.govtpgcollegeuna.in	-	-	-	-	-	-	60	-	-	-
29	L.R. Institute of Management, Village Jabli-Kyar, P.O. Oachghat, Tehsil Solan-173223 Phone No: 01792-252524 Website : www.lrinstitute.com	-	-	-	-	-	-	60	-	-	-
30	Himalayan College of Computer Science, Kala Amb, Tehsil Nahan, District Sirmour 173 030 (H.P.) Phone No: 01702-308500 to 308508 Website: www.himalayaninstitutions.com	-	-	-	-	-	-	60	-	-	-

Note :

1. Number of Engineering, Pharmacy and Management Institutions and their intake for the Session 2016-2017 may vary.
2. Admission of the candidates in the above mentioned institutions is subject to Affiliation, Continuation/Approval, timely deposit of relevant affiliation fee, inspection fee, other university fees and counselling fee etc. and satisfactory report submitted by the Inspection Committee constituted by H.P. Technical University for affiliation.



हिमाचल प्रदेश तकनीकी विश्वविद्यालय

हमीरपुर-177001, हिमाचल प्रदेश (भारत)

[हिमाचल प्रदेश अधिनियम-16, 2010 के अधीन स्थापित]

HIMACHAL PRADESH TECHNICAL UNIVERSITY

HAMIRPUR – 177001, HIMACHAL PRADESH (INDIA)

[ESTABLISHED UNDER STATE LAGISLATIVE ACT-16 OF 2010]

सूचना पत्रिका

प्रवेश प्रविवरण 2016-17

ADMISSION PROSPECTUS 2016-17

PART-II

(Will be uploaded on 1st June, 2016)

(APPLICABLE FOR ADMISSION TO VARIOUS TECHNICAL & PROFESSIONAL COURSES IN CAMPUS OF H.P. TECHNICAL UNIVERSITY AND ITS AFFILIATED COLLEGES ONLY)

IMPORTANT INSTRUCTIONS

APPLICANTS WILLING TO TAKE ADMISSION IN CAMPUS OF HIMACHAL PRADESH TECHNICAL UNIVERSITY AND ITS AFFILIATED COLLEGES/INSTITUTIONS

Applicants willing to take admission in Campus of Himachal Pradesh Technical University or colleges/Institutions affiliated to it are required to apply separately on the prescribed application form made available in its website. The Admission Prospectus containing instructions for filling online application-cum-counselling form and other related details shall be made available on the website of the University by June 1, 2016. The applicants have to attend centralized counselling to be conducted by this University as per schedule given in the Admission Prospectus.